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$$R^{m} \xrightarrow{R^{y_1}} O \xrightarrow{R^{x_1}} \stackrel{H}{R^{y_2}} = O$$

$$R^{m} \xrightarrow{N} R^{y_2} = O$$

$$R^{3a} \text{ (ii)}$$

$$-C \equiv C \xrightarrow{R^{3b}} R^{3b} \text{ (iii)}$$

$$R^{x_2} \xrightarrow{R^{y_3}} R^{y_3} \xrightarrow{R^{y_4}} R^{y_4}$$

$$R^{x_2} \xrightarrow{R^{y_3}} R^{y_4} \xrightarrow{R^{y_4}} R^{y_5} = O$$

(57) Abstract

Provided are pharmaceutical formulations, and methods of inhibiting fungal and parasitic activity using a compound of formula (I), C<sub>12</sub> alkyl) or -O-(CH<sub>2</sub>)<sub>q</sub>-X-R<sup>4</sup>, m is 2, 3 or 4; n is 2, 3, or 4; p is 0 or 1; q is 2, 3, or 4; X is pyrrolidino, piperidino or piperazino; and R<sup>4</sup> is hydrogen, C1-C12 alkyl, C3-C12 cycloalkyl, benzyl or C3-C12-cycloalkylmethyl; or a pharmaceutically acceptable salt thereof.

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### CYCLIC PEPTIDE ANTIFUNGAL AGENTS

This invention relates to semi-synthetic cyclic peptide compounds which are useful as antifungal and antiparasitic agents and which have improved stability and water solubility. In particular, it relates to derivatives of the echinocandin class of cyclic peptides; to methods for treating fungal and parasitic infections, and to formulations useful in the methods.

The compounds provided by this invention are semi-synthetic compounds derived from cyclic peptides which are produced by culturing various microorganisms. A number of cyclic peptides are known in the art including echinocandin B (A30912A), aculeacin, mulundocandin, sporiofungin, L-671,329, and S31794/F1.

In general, these cyclic peptides may be structurally characterized as a cyclic hexapeptide core (or nucleus) with an acylated amino group on one of the core amino acids. The amino group is typically acylated with a fatty acid group forming a side chain off the nucleus. For example, echinocandin B has a linoleoyl side chain while aculeacin has a palmitoyl side chain.

The fatty acid side chains may be removed from the cyclic peptide core to provide an amino nucleus (for example, a compound of formula I, below, where  $R_2$  is hydrogen). The amino group may then be re-acylated to provide semi-synthetic compounds such as those claimed in the present application.

The echinocandin B nucleus has been re-acylated with certain non-naturally occurring side chain moieties to provide a number of antifungal agents (see, <u>Debono</u>, U.S. Pat. No. 4,293,489). Among such antifungal agents is cilofungin which is represented by a compound of formula I where R', R", and R" are methyl; Rx1, Rx2, Ry1, Ry2, Ry3, Ry4 and R0 is hydroxy and R2 is p-(octyloxy)benzoyl.

The present invention provides a compound of the formula:

5 wherein:

R' is hydrogen, methyl or  $-CH_2C(O)NH_2$ ; R" and R" are independently methyl or hydrogen; R $^{x1}$  is hydrogen, hydroxy or -O-R;

R is  $C_1$ - $C_6$  alkyl, benzyl, -( $CH_2$ )<sub>2</sub>Si( $CH_3$ )<sub>3</sub>,

10 -CH<sub>2</sub>CHOHCH<sub>2</sub>OH, -CH<sub>2</sub>CH=CH<sub>2</sub>, -(CH<sub>2</sub>)<sub>a</sub>COOH, -(CH<sub>2</sub>)<sub>b</sub>NR<sup>z1</sup>R<sup>z2</sup>, -(CH<sub>2</sub>)<sub>c</sub>POR<sup>z3</sup>R<sup>z4</sup> or -[(CH<sub>2</sub>)<sub>2</sub>O]<sub>d</sub>-(C<sub>1</sub>-C<sub>6</sub>)alky1;

a, b and c are independently 1, 2, 3, 4, 5 or 6;  $\rm R^{z1}$  and  $\rm R^{z2}$  are independently hydrogen,  $\rm C_1\text{--}C_6$ 

alkyl, or  $R^{z1}$  and  $R^{z2}$  combine to form  $-CH_2(CH_2)_eCH_2$ ;

 $R^{z3}$  and  $R^{z4}$  are independently hydroxy or  $C_1$ - $C_6$ 

alkoxy;

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d is 1 or 2;

e is 1, 2 or 3;

 $R^{x2}$ ,  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$  and  $R^{y4}$  are independently

20 hydroxy or hydrogen;

 $R^0$  is hydroxy,  $-OP(O)(OH)_2$  or a group of the

formulae:

R<sup>1</sup> is C<sub>1</sub>-C<sub>6</sub> alkyl, phenyl, p-halo-phenyl, p-nitrophenyl, benzyl, p-halo-benzyl or p-nitro-benzyl;

$$R^2$$
 is  $-CH_2 \longrightarrow R^3$ ;

 $R^3$  is

$$\mathbb{R}^{3a}$$

$$C \equiv C - R^{3c}$$
 , or

$$\mathbb{R}^{3d}$$
 ;

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 $R^{3a}$ ,  $R^{3b}$ ,  $R^{3c}$  and  $R^{3d}$  are independently hydrogen,  $C_{1}$ - $C_{12}$  alkyl,  $C_{2}$ - $C_{12}$  alkynyl,  $C_{1}$ - $C_{12}$  alkoxy,  $C_{1}$ - $C_{12}$  alkylthio, halo, or -O- $(CH_{2})_{m}$ - $[O-(CH_{2})_{n}]_{p}$ -O- $(C_{1}$ - $C_{12}$  alkyl) or -O- $(CH_{2})_{q}$ -X- $R^{4}$ ;

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m is 2, 3 or 4;

n is 2, 3 or 4;

p is 0 or 1;

q is 2, 3 or 4;

X is pyrrolidino, piperidino or piperazino; and  $R^4$  is hydrogen,  $C_1$ - $C_{12}$  alkyl,  $C_3$ - $C_{12}$  cycloalkyl, benzyl or  $C_3$ - $C_{12}$  cycloalkylmethyl;

or a pharmaceutically acceptable salt thereof.

Also provided are pharmaceutical formulations, methods for inhibiting parasitic or fungal activity and methods of treating fungal or parasitic infections which employ the compounds of the invention.

As used herein, the term  $^{\circ}C_1-C_{12}$  alkyl $^{\circ}$  refers to a straight or branched alkyl chain having from one to twelve carbon atoms. Typical  $C_1-C_{12}$  alkyl groups include methyl, ethyl, propyl, isopropyl, butyl, sec-butyl, t-butyl, pentyl, 5-methylpentyl, hexyl, heptyl, 3,3-dimethylheptyl, octyl, 2-methyl-octyl, nonyl, decyl,

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undecyl, dodecyl and the like. The term "C1-C12 alkyl" includes within its definition the terms "C1-C6 alkyl" and C<sub>1</sub>-C<sub>4</sub> alkyl."

The term "C2-C12 alkynyl" refers to a straight or branched alkynyl chain having from two to twelve carbon atoms. Typical  $C_2$ - $C_{12}$  alkynyl groups include ethynyl, 1propyn-1-yl, 1-propyn-2-yl, 1-butyn-1-yl, 1-butyn-3-yl, 1pentyn-3-yl, 4-pentyn-2-yl, 1-hexyn-3-yl, 3-hexyn-1-yl, 5methyl-3-hexyn-1-yl, 5-octyn-1-yl, 7-octyn-1-yl, 4-decyn-1yl, 6-decyn-1-yl and the like.

The term "halo" refers to chloro, fluoro, bromo or iodo.

The term  ${}^{*}C_{1}-C_{12}$  alkylthio  ${}^{*}$  refers to a straight or branched alkyl chain having from one to twelve carbon atoms attached to a sulfur atom. Typical  $C_1$ - $C_{12}$  alkylthio groups include methylthio, ethylthio, propylthio, isopropylthio, butylthio, 3-methyl-heptylthio, octylthio, 5,5-dimethyl-hexylthio and the like.

The term "C1-C12 alkoxy" refers to a straight or branched alkyl chain having from one to twelve carbon atoms attached to an oxygen atom. Typical  $C_1$ - $C_{12}$  alkoxy groups include methoxy, ethoxy, propoxy, butoxy, sec-butoxy, pentoxy, 5-methyl-hexoxy, heptoxy, octyloxy, decyloxy dodecyloxy and the like. The term "C1-C12 alkyl" includes 25 · within its definition the terms "C1-C6 alkoxy" and C1-C4 alkoxy."

> The term "C3-C12 cycloalkyl" refers a saturated hydrocarbon ring structure having from three to twelve carbon atoms. Typical C3-C12 cycloalkyl groups include cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and cycloheptyl, cyclooctyl and the like.

The term "hydroxy protecting group" refers to a substituent of an hydroxy group that is commonly employed to block or protect the hydroxy functionality while reactions are carried out on other functional groups on the compound. Examples of such hydroxy protecting groups include tetrahydropyranyl, 2-methoxyprop-2-yl, 1-ethoxyeth-

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1-yl, methoxymethyl,  $\beta$ -methoxymethyl, methylthiomethyl, t-butyl, t-amyl, trityl, 4-methoxytrityl, 4,4'-dimethoxytrityl, 4,4',4"-trimethoxytrityl, benzyl, allyl, trimethylsilyl, (t-butyl)dimethylsilyl, and 2,2,2trichloroethoxycarbonyl and the like. The species of 5 hydroxy protecting group is not critical so long as the derivatized hydroxy group is stable to the conditions of the subsequent reaction(s) and can be removed at the appropriate point without disrupting the remainder of the molecule. A preferred hydroxy protecting group is 10 trimethylsilyl. Further examples of hydroxy protecting groups are described in T.W. Greene, "Protective Groups in Organic Synthesis, " John Wiley and Sons, New York, N.Y., (2nd ed., 1991) chapters 2 and 3. The term "protected hydroxy" refers to a hydroxy group bonded to one of the 15 above hydroxy protecting groups.

The term "amino protecting group" as used in the specification refers to substituents of the amino group commonly employed to block or protect the amino functionality while reacting other functional groups on the compound. Examples of such amino protecting groups include formyl, trityl, phthalimido, trichloroacetyl, chloroacetyl, bromoacetyl, iodoacetyl groups, or urethane-type blocking groups such as benzyloxycarbonyl, 4-phenylbenzyloxycarbonyl, 2-methylbenzyloxycarbonyl, 4-methoxybenzyloxycarbonyl, 4-fluorobenzyloxycarbonyl, 4-chlorobenzyloxycarbonyl, 3-chlorobenzyloxycarbonyl, 2-chlorobenzyloxycarbonyl, 2,4-dichlorobenzyloxycarbonyl, 4-bromobenzyloxycarbonyl, 3-bromobenzyloxycarbonyl, 4-

nitrobenzyloxycarbonyl, 4-cyanobenzyloxycarbonyl, t-butoxycarbonyl, 2-(4-xenyl)isopropoxycarbonyl, 1,1-diphenyleth-1-yloxycarbonyl, 1,1-diphenylprop-1-yloxycarbonyl, 2-phenylprop-2-yloxycarbonyl, 2-(p-toluyl)-prop-2-yloxycarbonyl, cyclopentanyloxycarbonyl, 1-

methylcyclopentanyloxycarbonyl, cyclohexanyloxycarbonyl, 1methylcyclohexanyloxycarbonyl, 2methylcyclohexanyloxycarbonyl, 2-(4-toluylsulfonyl)-

1981, Chapter 7.

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ethoxycarbonyl, 2-(methylsulfonyl)ethoxycarbonyl, 2-(triphenylphosphino)-ethoxycarbonyl, fluorenylmethoxycarbonyl ("FMOC"), 2-(trimethylsilyl)ethoxycarbonyl, allyloxycarbonyl, 1-(trimethylsilylmethyl)prop-1-5 enyloxycarbonyl, 5-benzisoxalylmethoxycarbonyl, 4acetoxybenzyloxycarbonyl, 2,2,2-trichloroethoxycarbonyl, 2ethynyl-2-propoxycarbonyl, cyclopropylmethoxycarbonyl, 4-(decyloxy) benzyloxycarbonyl, isobornyloxycarbonyl, 1piperidyloxycarbonyl and the like; benzoylmethylsulfonyl, 10 2-nitrophenylsulfenyl, diphenylphosphine oxide and like amino protecting groups. The species of amino protecting group employed is not critical so long as the derivatized amino group is stable to the condition of subsequent reaction(s) on other positions of the intermediate molecule and can be selectively removed at the appropriate point 15 without disrupting the remainder of the molecule including any other amino protecting group(s). Preferred amino protecting groups are t-butoxycarbonyl (t-Boc), allyloxycarbonyl and benzyloxycarbonyl (CbZ). Further 20 examples of groups referred to by the above terms are described by J. W. Barton, "Protective Groups in Organic Chemistry", J. G. W. McOmie, Ed., Plenum Press, New York, N.Y., 1973, Chapter 2, and T. W. Greene, "Protective Groups in Organic Synthesis", John Wiley and sons, New York, N.Y., **25** .

> The term "inhibiting", i.e. a method of inhibiting parasitic or fungal activity, includes stopping, retarding or prophylactically hindering or preventing the growth or any attending characteristics and results from the existence of a parasite or fungus.

> The term "contacting", i.e. contacting a compound of the invention with a parasite or fungus, includes a union or junction, or apparent touching or mutual tangency of a compound of the invention with a parasite or fungus. However, the term does not imply any further limitations to the process, such as by mechanism of inhibition, and the methods are defined to encompass the

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spirit of the invention, which is to inhibit parasitic and fungal activity by the action of the compounds and their inherent antiparasitic and antifungal properties, or in other words, the compounds, used in the claimed methods are the causative agent for such inhibition.

The term "pharmaceutically acceptable salt" as used herein, refers to salts of the compounds of the above formula which are substantially non-toxic to living organisms. Typical pharmaceutically acceptable salts include those salts prepared by reaction of the compounds of the present invention with a mineral or organic acid or an inorganic base. Such salts are known as acid addition and base addition salts.

Acids commonly employed to form acid addition 15 salts are mineral acids such as hydrochloric acid, hydrobromic acid, hydroiodic acid, sulfuric acid, phosphoric acid and the like, and organic acids such as ptoluenesulfonic, methanesulfonic acid, oxalic acid, pbromophenylsulfonic acid, carbonic acid, succinic acid, 20 citric acid, benzoic acid, acetic acid, and the like. Examples of such pharmaceutically acceptable salts are the sulfate, pyrosulfate, bisulfate, sulfite, bisulfite, phosphate, monohydrogenphosphate, dihydrogenphosphate, metaphosphate, pyrophosphate, chloride, bromide, iodide, acetate, propionate, decanoate, caprylate, acrylate, 25 formate, isobutyrate, caproate, heptanoate, propiolate, oxalate, malonate, succinate, suberate, sebacate, fumarate, maleate, butyne-1,4-dioate, hexyne-1,6-dioate, benzoate, chlorobenzoate, methylbenzoate, dinitrobenzoate, 30 hydroxybenzoate, methoxybenzoate, phthalate, sulfonate, xylenesulfonate, phenylacetate, phenylpropionate, phenylbutyrate, citrate, lactate,  $\gamma$ -hydroxybutyrate, glycollate, tartrate, methanesulfonate, propanesulfonate, naphthalene-1-sulfonate, napththalene-2-sulfonate, mandelate and the like. Preferred pharmaceutically 35 acceptable acid addition salts are those formed with mineral acids such as hydrochloric acid and hydrobromic

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acid, and those formed with organic acids such as maleic acid and methanesulfonic acid.

Base addition salts include those derived from inorganic bases, such as ammonium or alkali or alkaline earth metal hydroxides, carbonates, bicarbonates, and the like. Such bases useful in preparing the salts of this invention thus include sodium hydroxide, potassium hydroxide, ammonium hydroxide, potassium carbonate, sodium carbonate, sodium bicarbonate, potassium bicarbonate, calcium hydroxide, calcium carbonate, and the like. The potassium and sodium salt forms are particularly preferred.

It should be recognized that the particular counterion forming a part of any salt of this invention is not of a critical nature, so long as the salt as a whole is pharmacologically acceptable and as long as the counterion does not contribute undesired qualities to the salt as a whole.

Preferred compounds of this invention are those compounds of formula I where:

20 R', R" and R" are each methyl;

 $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$  and  $R^{y4}$  are each hydroxy;

Rx1 is hydrogen, hydroxy or -O-R;

R is methyl, benzyl, -CH2CHOHCH2OH, -(CH2)bNRz1Rz2 or -(CH2)2PORz3Rz4;

b is 2, 3, 4, 5 or 6;

 $\mbox{R}^{z1}$  and  $\mbox{R}^{z2}$  are independently hydrogen or  $\mbox{C}_1\mbox{-}\mbox{C}_4$ 

alkyl;

 $R^{z3}$  and  $R^{z4}$  are independently hydroxy or methoxy;  $R^{x2}$  is hydrogen or hydroxy;

R<sup>0</sup> is hydroxy, or a group of the formulae:

R<sup>1</sup> is methyl;

or a pharmaceutically acceptable salt thereof.

Of these compounds, more preferred are those compounds of formula I where:

R<sup>x1</sup> is hydroxy;

R<sup>x2</sup> is hydroxy;

R<sup>0</sup> is hydroxy;

 $\mathbb{R}^2$  is a group of the formula:

$$-$$
C $\equiv$ C $-$ R $^{3c}$  , or

$$R^{3d}$$

 $\rm R^{3c}$  and  $\rm R^{3d}$  are independently hydrogen,  $\rm C_2-C_{12}$  alkynyl,  $\rm C_1-C_{12}$  alkoxy or -O-(CH<sub>2</sub>)<sub>m</sub>-[O-(CH<sub>2</sub>)<sub>n</sub>]<sub>p</sub>-O-(C<sub>1</sub>-C<sub>12</sub> alkyl); or a pharmaceutically acceptable salt thereof.

Of these compounds, the most preferred are those compounds where

$$R^2$$
 is  $-CH_2$ 

or

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$$R^2$$
 is  $-CH_2$ —C $\equiv C$ 

or a pharmaceutically acceptable salt thereof.

The compounds of formula I may be prepared

The compounds of formula I may be prepared as follows:

### Reaction Scheme I

wherein:

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 $${\tt R}^{\tt nat}$$  is a naturally occurring cyclic peptide sidechain; and

R', R", R", Rx1, Rx2, Ry1, Ry2, Ry3, Ry4, R0 and  $\mbox{R}^2$  are as defined above.

Reaction scheme I, above, is accomplished by carrying out reactions A-C, in order. Once a reaction is complete, the intermediate compound may be isolated by procedures well-known in the art, for example, the compound may be crystallized or precipitated and then collected by filtration, or the reaction solvent may be removed by extraction, evaporation or decantation. The intermediate compound may be further purified, if desired, by common techniques such as crystallization or precipitation, or chromatography over solid supports such as silica gel, alumina and the like, before carrying out the next step of the reaction scheme.

In reaction IA, a naturally occurring cyclic peptide of the formula IA is deacylated using procedures known in the art to provide an amino nucleus of formula IB. This reaction is typically carried out using enzymatic deacylation by exposing the naturally occurring cyclic peptide to a deacylase enzyme. The deacylase enzyme may be obtained from the microorganism *Actinoplanes utahensis* and

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used substantially as described in U.S. Patent Nos. 4,293,482 and 4,304,716, herein incorporated by reference. The deacylase enzyme may also be obtained from the Pseudomonas species. Deacylation may be accomplished using whole cells of Actinoplanes utahensis or Pseudomonas or the crude or purified enzyme thereof or using an immobilized form of the enzyme. See European Patent Application No. 0460 882 (December 11, 1991). Examples of naturally occurring cyclic peptides which may be used as starting materials include aculeacin (palmitoyl side chain), tetrahydroechinocandin B (stearoyl side chain), mulundocandin (branched  $C_{15}$  side chain), L-671,329 ( $C_{16}$  branched side chain), S 31794/F1 (tetradecanoyl side chain), sporiofungin ( $C_{15}$  branched side chain), FR901379 (palmitoyl side chain) and the like. A preferred naturally occurring cyclic peptide is echinocandin B (a compound of formula IA where R', R" and R" are each methyl,  $R^{x1}$ ,  $R^{x2}$ ,  $\mathbb{R}^{y1}$ ,  $\mathbb{R}^{y2}$ ,  $\mathbb{R}^{y3}$ ,  $\mathbb{R}^{y4}$  and  $\mathbb{R}^{0}$  are each hydroxy and  $\mathbb{R}^{2}$  is linoleov1).

In Reaction IB, the resulting amino nucleus is N-alkylated using reductive amination to provide a compound of formula I where  $R^2$  is as defined hereinabove. reaction is typically carried out by reacting the amino nucleus of formula IB with an appropriately substituted  $\cdot$ aldehyde of the formula  $R^2$ -COH in the presence of a reducing agent such as sodium cyanoborohydride. reaction is typically carried out for one to sixty five hours at a temperature of from about 20°C to about 100°C in a mutual inert solvent. Typical solvents for this reaction include dimethylformamide, methanol or a mixture of such solvents. Solvent choice is not critical so long as the solvent employed is inert to the ongoing reaction and the reactants are sufficiently solubilized to effect the desired reaction. The aldehyde reactant is generally employed in a slight excess relative to the amino nucleus.

The compounds of formula I where  $\mathbb{R}^{\times 1}$  is hydroxy may be reacted with an appropriately substituted alcohol in

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the presence of an acid to provide a compound of formula I where R<sup>x1</sup> is -O-R, where R is C<sub>1</sub>-C<sub>6</sub> alkyl, benzyl, -(CH<sub>2</sub>)<sub>2</sub>Si(CH<sub>3</sub>)<sub>3</sub>, -CH<sub>2</sub>CH=CH<sub>2</sub>, -(CH<sub>2</sub>)<sub>a</sub>COOH, -(CH<sub>2</sub>)<sub>b</sub>NR<sup>z1</sup>R<sup>z2</sup>, -(CH<sub>2</sub>)<sub>c</sub>POR<sup>z3</sup>R<sup>z4</sup> or -[(CH<sub>2</sub>)<sub>2</sub>O]<sub>d</sub>-(C<sub>1</sub>-C<sub>6</sub>)alkyl. The reaction is typically carried out in a polar aprotic solvent such as dioxane or dimethylsulfoxide at a temperature of from about 0°C to about 35°C, preferably at about room temperature. Solvent choice is not critical so long as the solvent employed is inert to the ongoing reaction and the reactants are sufficiently solubilized to effect the desired reaction. Preferred acids include p-toluenesulfonic acid, hydrochloric acid and camphorsulfonic acid.

The compounds of formula I where  $R^{x1}$  is  $-(CH_2)_bNR^{z1}R^{z2}$  where  $R^{z1}$  and  $R^{z2}$  are hydrogen may be prepared via a protected compound wherein  $R^{x1}$  is  $-(CH_2)_bNHR^a$  where  $R^a$  is an amino protecting group. The resultant protected compound is then deprotected according to procedures known in the art.

The compounds of formula I where Rx1 is -CH<sub>2</sub>CHOHCH<sub>2</sub>OH may be prepared by hydroxylating a compound of formula I where  $R^{\mathbf{x}\mathbf{1}}$  is  $-CH_2CH=CH_2$  with osmium tetroxide in the presence of a catalyst at a temperature in the range of from about 0°C to about 40°C for about one to twenty four hours in a organic/aqueous solvent mixture, for example dioxane/water. Suitable catalysts include Nmethylmorpholine N-oxide (NMO) and the like. Typical solvents suitable for use in this reaction include dimethylformamide, tetrahydrofuran, acetone and dioxane. Solvent choice is not critical so long as the solvent employed is inert to the ongoing reaction and the reactants are sufficiently solubilized to effect the desired reaction. The reaction is preferably conducted at a temperature in the range of from about 20°C to about 30°C for about eighteen to twenty four hours.

The compounds of formula I where  $\mathbb{R}^0$  is hydroxy may be phosphorylated by reaction with an appropriately substituted alkyl or phenyl phosphate to provide a compound

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of formula I where  $R^0$  is  $-0-P(0)OH-R^1$  where  $R^1$  is  $C_1-C_6$ alkoxy or phenoxy, or by reaction with an appropriately substituted alkyl or phenyl phosphonic acid to provide a compound of formula I where  $R^0$  is  $-O-P(O)OH-R^1$  where  $R^1$  is  $C_1$ - $C_6$  alkyl, or an appropriately substituted phenyl or benzyl moiety, to provide a compound of formula I where  ${\ensuremath{\mathtt{R}}}^0$ is a group of the formula  $-OP(O)OH-R^{1}$ . The phosphonic acid is typically used in an activated form, for example as a phosphonic halide, preferably a phosphonic chloride. The reaction is carried out in the presence of a base such as lithium trimethylsilanolate (LiOTMS), lithium bis(trimethylsilyl)amide (LHMDS), pyridine and the like. The reaction is typically carried out for up to one hour at a temperature from about -30°C to about 0°C in an aprotic solvent such as tetrahydrofuran and dimethylformamide. The reaction is generally complete in about fifteen minutes when carried out under these conditions. The phosphate or phosphonate reactant is generally employed in equimolar proportions to about a one mole excess relative to the amino nucleus in the presence of an equimolar or slight excess of the base. Phosphorylation of an amino nucleus with unprotected aminal hydroxy groups is typically carried out at lower temperatures, for example from about -30°C to about -15°C.

Alternatively, the aminal hydroxy moieties on the compound of formula I are optionally protected with an hydroxy protecting group using procedures known in the art. For example, the reaction is typically carried out by combining the compound of formula I with a suitable hydroxy protecting group in the presence of a catalyst at a temperature in the range of from about 0°C to about 40°C for about one to five hours in a mutual inert solvent. The hydroxy protecting group is generally employed in an amount ranging from about equimolar proportions to about a 100 molar excess relative to the compound of formula I, preferably in a large molar excess. Suitable catalysts include strong acids such as p-toluenesulfonic acid,

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camphorsulfonic acid (CSA), hydrochloric acid, sulfuric acid, trifluoroacetic acid and the like. Typical solvents suitable for use in this reaction include any organic solvent such as dioxane. Solvent choice is not critical so long as the solvent employed is inert to the ongoing reaction and the reactants are sufficiently solubilized to effect the desired reaction. The reaction is preferably conducted at a temperature in the range of from about 20°C to about 30°C for about two to four hours. The protected compound of formula I is then phosphorylated as described above. The hydroxy protecting group(s) are then removed according to procedures known in the art to provide a phosphorylated compound of formula I. For example, the protecting groups can be removed by reaction with a Lewis acid in a mutual inert organic solvent such as methylene chloride. Examples of Lewis acids include trimethylsilylbromide, boron trifluoride etherate and the like. The reaction is typically carried out at a temperature of from about 0°C to about 40°C, preferably at a temperature of from about 20°C to about 30°C. A preferred Lewis acid is boron trifluoride etherate.

The dideoxy compounds of formula I are prepared by removing the benzylic and aminal hydroxy groups (Rx2 and Rx1, respectively). The hydroxy groups may be removed by subjecting a non-dideoxy compound of formula I (where  $R_2$  is hydrogen or acyl) to a strong acid and a reducing agent at a temperature of between -5°C and 70°C, in a suitable solvent. Typical strong acids include trichloroacetic acid, trifluoroacetic acid or borontrifluoride etherate. preferred strong acid is trifluoroacetic acid. Typical reducing agents include sodium cyanoborohydride or triethylsilane. A preferred reducing agent is triethylsilane. Suitable solvents include methylene chloride, chloroform or acetic acid, preferably methylene chloride. The strong acid should be present in an amount of from 2 to 80 mol per mol of substrate, and the reducing agent should be present in an amount of 2 to 80 mol per mol

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of substrate. This process affords selective removal of the aminal and benzylic hydroxy groups.

The cyclic peptides used to make the compounds of the present invention may be prepared by fermentation of known microorganisms. For example, the cyclic peptide of formula IB where R', R" and R" are methyl,  $R^{x1}$ ,  $R^{x2}$ ,  $R^{y1}$ ,  ${\rm R}^{y2},~{\rm R}^{y3},~{\rm R}^{y4}$  are hydroxy and  ${\rm R}^0$  is hydroxy (cyclic nucleus corresponding to A-30912A) may be prepared using the procedure detailed in <u>Abbott et al.</u>, U.S. Pat. Ser. No. 4,293,482, which is herein incorporated by reference. cyclic peptide of formula IB where R', R" and R"" are methyl,  $R^{x1}$  is hydroxy,  $R^{x2}$  is hydrogen,  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$ ,  $R^{y4}$ and  ${\ensuremath{\mathsf{R}}}^0$  is hydroxy (cyclic nucleus corresponding to A-30912B) may be prepared using the procedure detailed in Abbott et al., U.S. Pat. Ser. No. 4,299,763, which is herein incorporated by reference. Aculeacin may be prepared using the procedure detailed in Mizuno et al., U.S. Pat. Ser. No. 3,978,210 which is herein incorporated by reference. The cyclic peptide of formula IB where R' is -CH<sub>2</sub>C(O)NH<sub>2</sub>, R" is methyl, R" is hydrogen,  $R^{\times 1}$ ,  $R^{\times 2}$ ,  $R^{y1}$ ,  $\mathbb{R}^{y2}$ ,  $\mathbb{R}^{y3}$ ,  $\mathbb{R}^{y4}$  and  $\mathbb{R}^{0}$  is hydroxy may be prepared by deacylating the cyclic peptide prepared using the procedure detailed in Chen et al., U.S. Pat. Ser. No. 5,198,421,

which is herein incorporated by reference.

The aldehydes of the formula R<sup>2</sup>-COH, used in the reductive aminiation, may be obtained commercially or prepared according to procedures known in the art. For example, an appropriately substituted phenyl boronic acid or biphenyl boronic acid reactant may be reacted with a phalobenzaldehyde reactant in the presence of a catalyst such as tetrakis(triphenylphosphine)palladium and an inorganic base such as potassium carbonate in a mutual inert organic solvent such as toluene at a temperature of from about 20°C to the reflux temperature of the reaction mixture to provide the corresponding biphenyl aldehydes and terphenyl aldehydes used to prepare the compounds of formula I. The reaction is typically carried out with

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equimolar proportions of the boronic acid reactant and the p-benzaldehyde reactant, or a slight molar excess of the p-benzaldehyde reactant relative to the boronic acid reactant, and a 1-2 molar excess of the inorganic base. The reaction is generally complete after about four to about ten hours when carried out at reflux temperature in toluene.

The boronic acid reactant may be prepared by reacting an appropriately substituted halophenyl or halobiphenyl reactant with two equivalents of triisopropyl borate in the presence of an alkyl lithium, for example sec-butyl lithium, in a mutual inert solvent such as tetrahydrofuran. The alkyl lithium is typically employed in a slight molar excess relative to the halophenyl or halobiphenyl reactant. The alkyl lithium is typically combined with the solvent by dropwise addition at reduced temperatures (<-70°C) and allowed to stir for approximately thirty minutes before the addition of the triisopropyl borate. The reaction is typically carried out initially at a temperature of from about -100°C to about -50°C, preferably from about -75°C to about -85°C for thirty minutes to two hours and then warmed to room temperature and reacted for an additional one to three hours. reaction is generally complete in from several minutes to about four hours. When the reaction is substantially complete, the boronic acid moiety is formed by the addition of an acid. A preferred acid is a 1N hydrochloric acid solution.

The R<sup>2</sup>-COH aldehydes having an acetylene moiety

may be prepared by reacting an appropriately substituted
acetylene reactant with an appropriately substituted phenyl
or biphenyl reactant of the formula

where L is a suitable leaving group such as bromo, iodo,
methanesulfonate, toluenesulfonate,
trifluoromethanesulfonate and the like, in the presence of

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a catalyst and preferably in the presence of an acid scavenger in a mutual inert solvent such as acetonitrile. Examples of acid scavengers include triethylamine and pyridine, preferably triethylamine. A preferred catalyst is formed in situ from palladium (II) chloride, triphenylphosphine and copper (I) iodide. The reaction is typically carried out for thirty minutes to twenty one hours at a temperature from about room temperature to the reflux temperature of reaction mixture. The reaction is generally complete after about two to about six hours when carried out at reflux temperature.

Alternatively, a suitably substituted phenyl reactant of the formula halo—L may be reacted with an appropriately substituted acetylene reactant as described above to provide, for example, a compound of the formula halo— $C\equiv C$ — $R^{3c}$  which can be coupled with a phenyl boronic acid reactant as described above.

describe how to synthesize the compounds of the present invention. The terms melting point, proton nuclear magnetic resonance spectra, mass spectra, infrared spectra, ultraviolet spectra, elemental analysis, high performance liquid chromatography, and thin layer chromatography are abbreviated "m.p.", "NMR", "MS", "IR", "UV", "Analysis", "HPLC", and "TLC", respectively. In addition, the absorption maxima listed for the IR spectra are only those of interest and not all of the maxima observed.

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### <u>Preparation 1</u> 4-Octyloxybenzaldehyde

A solution containing 3.053 g (25 mmol) of 4-formylphenol, 6.48 ml (3705 mmol) of 1-bromooctane and 6.9 mg (50 mmol) of potassium carbonate in 100 ml of acetone was refluxed overnight. When the reaction was substantially complete, as indicated by thin layer

chromatagraphy (TLC), the reaction was quenched by the addition of 100 ml of water. The desired compound was extracted from the reaction mixture using two 100 ml portions of diethyl ether. The resultant solution was dried over magnesium sulfate, filtered and then concentrated in vacuo to provide a liquid which was purified using HPLC (eluent of 10 ethyl acetate in hexane) to provide the desired compound.

MS(FAB): 235.2 (M+H).

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## Preparation 2 A. (D) B (OH) 2

To a cold (-78C) solution of 10.0 mg (42.9 mmol) of 1-bromo-4-phenylbenzene, was added 42.9 ml of a  $1.3\underline{M}$ solution of sec-butyllithium in tetrahydrofuran (55.8 15 mmol), dropwise. To the resultant mixture was added 14.85 ml (64.35 mmol) of triisopropyl borate, dropwise. resultant reaction mixture was stirred for approximately thirty minutes and then warmed to room temperature and 20 allowed to react for approximately two hours. was then quenched by the addition of approximately 50 ml of  $1\underline{\mathrm{N}}$  hydrochloric acid and the resultant mixture was concentrated in vacuo to provide a residue. was redissolved in diethyl ether, filterd and dried in vacuo to provide 1.58 g of the desired subtitled compound. 25

A 2M solution of sodium carbonate was added to a solution of 2.970 g (15 mmol) of the compound of

Preparation 2A in 120 ml of toluene. After degassing the resultant mixture, 3.470 g (18.75 mmol) of 1-bromo-4-formylbenzene and 1.713 g (1.5 mmol) of tetrakis(triphenylphosphine)palladium were added to the above solution and the resultant reaction mixture was refluxed overnight. When the reaction was substantially

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complete, as indicated by TLC, the reaction mixture was cooled to room temperature and concentrated *in vacuo* to provide a residue. This residue was redissolved in methylene chloride and washed with two 30 ml portions of brine. The organic portion was then filtered and dried *in vacuo* to provide a solid.

## A. Br-C-C-O(CH<sub>2</sub>)<sub>4</sub>CH<sub>3</sub>

10 A solution containing 50 g (200 mmol) of 4bromophenol, 33.5 g (298 mmol) of potassium t-butoxide and 40 ml (298 mmol) of 1-iodopentane in 1000 ml of tetrahydrofuran was reacted at reflux temperature for approximately twenty four hours. When the reaction was 15 substantially complete, as indicated by TLC, the reaction was filtered. The resultant filtrate was concentrated in vacuo to provide a purple solid. This solid was redissolved in a water/diethyl ether mixture to provide a yellow solution. This solution was washed sequentially 20 with 200 ml of water (twice), 100 ml of 2N sodium hydroxide (twice) and 200 ml of brine (twice), dried over sodium sulfate and then concentrated in vacuo to provide a yellow powder. This solid was recrystallized from hot hexanes to provide a white powder. 25 Yield: 45.8 mg (72%).

B. 
$$(OH)_2B$$
  $O(CH_2)_4CH_3$ 

To a cold (-78°C) solution of 10.0 mg (42.9 mmol) of 29 g (90.8 mmol) of the compound of Preparation 1A, was added 91 ml of sec-butyllithium in 1000 ml of tetrahydrofuran (118 mmol), dropwise. To the resulting mixture was added 41.9 ml (181.7 mmol) of triisopropyl borate, dropwise. The resultant reaction mixture was stirred for approximately thirty minutes and then warmed to room temperature and allowed to react for approximately two

hours. The reaction was then quenched by the addition of 1N hydrochloric acid. The resultant mixture was concentrated *in vacuo* to provide a residue. This residue was redissolved in diethyl ether, filtered and dried to provide the desired subtitled compound. Yield:

A solution of 4.87 mg (26.2 mmol) of 1-bromo-4-10 formyl benzene in methanol was added to a solution containing 6 g (21 mmol) of the compound of Preparation 3B, 60 ml of 2M sodium carbonate and 2.5 g (2.1 mmol) of tetrakis(triphenylphosphine)palladium in 120 ml of toluene. The resultant reaction mixture was allowed to react at 15 reflux temperature for approximately five hours. When the reaction was substantially complete, as indicated by TLC, the biphasic mixture was separated and the organic layer was washed sequentially with water and brine, dried over magnesium sulfate, filtered and concentrated in vacuo to 20 provide a solid. This solid was recrystallized from hot hexanes.

MS(FD): 344(M+).

# Preparation 4 O H-C-C-C-C-C

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To a solution of 2.5 g (13 mmol) of p-bromo benzaldehyde in 16 ml of acetonitrile, was added 1.5 g (14 mmol) of phenyl acetylene, 0.55 g (0.52 mmol) of palladium-on-copper, 0.54 g (2 mmol) of triphenylphosphine, 0.1 g (0.52 mmol) of copper (I) iodide and 32.5 ml of triethylamine. The resultant reaction mixture was degassed in vacuo and flushed with argon (three times). After the reaction mixture was refluxed, under argon, for twenty four hours, the mixure was cooled to room temperature and

concentrated *in vacuo* to provide a residue. This residue was purified using flash chromatography (silica gel; eluent of 20% ethyl acetete in hexanes) to provide 1 g of a white powder.

5 Yield: 37%.

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):

 $\delta$  7.4 (m, 3H), 7.6 (m, 2H), 7.7, (d, J=7.68 Hz, 2H), 7.85 (d, J=7.68 Hz, 2H), 10.02 (s, 1H).

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## Preparation 5 A. Br-C=C=C

To a solution containing 5 g (21.2 mmol) of 1,4-dibromobenzene, 18.8 mg (0.106 mmol) of palladium (II) chloride, 55.6 mg (0.212 mmol) of triphenylphosphine and 5.91 ml (0.726 mmol) of triethylamine in 300 ml of acetonitrile, was added 2.327 g (21.2 mmol) of phenyl acetylene and 40.0 mg (0.212 mmol) of copper (I) iodide. The resultant reaction mixture was allowed to react at room temperature for approximately two days. The crude material was purified using HPLC (eluent of hexane) to provide 660 mg of a white solid.

## B. H-Ü-(C)-(C)-(C)-(C)

The desired subtitled compound was prepared

substantially in accordance with the procedure detailed in Preparation 2B, using 3.07 g (11.9 mmol) of the subtitled compound of Preparation 5A and 1.78 g (11.9 mmol) of 1-boronic acid-4-formylbenzene, 60 ml of 2M sodium carbonate and 1.360 g (1.19 mmol) of tetrakis(triphenylphosphine)

palladium in 90 ml of toluene.

MS(FAB): 283.1 (M+H).

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#### Example 1

Preparation of the compound of formula I where R'. R" and R" are each methyl,  $R^{x1}$ ,  $R^{x2}$ ,  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$ ,  $R^{y4}$  and  $R^0$  are each hydroxy and  $R^2$  is 4-octyloxybenzyl

A solution containing 1.5 g (1.88 mmol) of the (A-30912A) nucleus (compound of formula IB where R', R" and R" are each methyl, Rx1, Rx2, Ry1, Ry2, Ry3, Ry4 and R0 are each hydroxy), 697 mg (2.07 mmol) of the compound of Preparation 1, and 130 mg (2.07 mmol) of sodium cyanoborohydride in a 1:1 dimethylformamide/methanol mixture was heated at 70°C overnight. When the reaction was substantially complete, as indicated by TLC, the reaction mixture was concentrated in vacuo the desired compound was isolated using HPLC (eluent of 40% aqueous acetonitrile; 60 ml/min.; 280 nm). The fractions containing the desired compound were combined and concentrated in vacuo to provide crude material. This material was purified using HPLC (eluent of 50% aqueous acetonitrile; 50 ml/min.; 280 nm).

Yield: 19 mg.

MS(FAB) for  $C_{49}H_{72}N_{7}O_{15}$ :

Calcd: 998.5086 (M-H<sub>2</sub>O);

Found: 998.5076.

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#### Example 2

Preparation of the compound of formula I where R'. R" and R" are each methyl, R\*1, R\*2, R\*1, R\*2, R\*2, R\*3, R\*4

and R<sup>0</sup> are each hydroxy and

 $R^2$  is  $-CH_2$ 

The desired compound was prepared substantially in accordance with the procedure detailed in Example 1 using 1.5 g (1.88 mmol) of the (A-30912A) nucleus (compound of formula IB where R', R" and R" are each methyl,  $R^{x1}$ ,  $R^{x2}$ ,  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$ ,  $R^{y4}$  and  $R^{0}$  is hydroxy), 533.5 mg (2.068 mmol) of the compound of Preparation 2B, and 130 mg (2.07 mmol) of sodium cyanoborohydride in 100 ml of a 1:1

dimethylformamide/methanol mixture with the exception that the reaction was substantially complete after approximately twelve hours. The crude material was purified using HPLC (eluent of 50% aqueous acetonitrile; 60 ml/min.; 280 nm).

5 Yield: 24 mg.

MS(FAB) for C<sub>53</sub>H<sub>65</sub>N<sub>7</sub>O<sub>15</sub>:

Calcd: 1040.4617 (M+H);

Found: 1040.4636.

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#### Example 3

Preparation of the compound of formula I where R', R" and R" are each methyl, R\*1, R\*2, R\*1, R\*2, R\*2, R\*3, R\*4

and R\*1 are each hydroxy and

 $R^2$  is  $-CH_2$   $O(CH_2)_4CH_3$ 

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The desired compound was prepared substantially in accordance with the procedure detailed in Example 1 using 1 g (1.25 mmol) of the (A-30912A) nucleus (compound of formula IB where R', R" and R" are each methyl, Rxl, Rx2, Ryl, Ry2, Ry3, Ry4 and R0 is hydroxy), 474.0 mg (1.38 mmol) of the compound of Preparation 3C, and 86.7 mg (1.38 mmol) of sodium cyanoborohydride in 100 ml of a 3:1 methanol/dimethylformamide mixture, with the exception that the reaction was substantially complete after approximately six hours. After isolating the crude material using HPLC (eluent of 50% aqueous acetonitrile; 60 ml/min.; 280 nm), the fractions containing the desired compound were combined, concentrated in vacuo and lyophilized.
MS(FAB): 1132.5 (M+Li).

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#### Example 4A

Preparation of the compound of formula I where R', R" and R" are each methyl, Rx2, Ry1, Ry2, Ry3, Ry4 and R0 are each hydroxy, Rx1 is hydrogen, and

 $R^2$  is  $-CH_2$ 

A solution of 203.0 mg (0.253 mmol) of the (A-30912A) nucleus (compound of formula IB where R', R" and R" are each methyl,  $R^{x1}$ ,  $R^{x2}$ ,  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$ ,  $R^{y4}$  and  $R^{0}$  is hydroxy) and 83.0 mg (0.455 mmol) of 4-phenylbenzaldehyde in 10 ml of methanol was reacted at reflux temperature. Yield: 22 mg.

MS(FAB) for C47H61N7O15:

Calcd: 964.4348 (M+H);

Found: 964.4304.

#### Example 4B

Alternate Preparation of the compound of formula I where R'. R" and R" are each methyl. Rx1, Rx2, Ry1, Ry2, Ry3, Ry4 and R<sup>0</sup> are each hydroxy and

$$R^2$$
 is  $-CH_2$ 

A solution of 1.5 g (1.88 mmol) of the (A-30912A) nucleus (compound of formula IB where R', R" and R" are each methyl, Rx1, Rx2, Ry1, Ry2, Ry3, Ry4 and R<sup>0</sup> are each hydroxy), 376.8 mg (2.068 mmol) of 4-phenylbenzaldehyde and 130 mg (2.07 mmol) of sodium cyanoborohydride in a 100 ml of a 3:1 methanol/dimethylformamide mixture was allowed to react overnight at reflux temperature. The resultant crude material was isolated using HPLC (eluent of 50% aqueous acetonitrile; 60 ml/min.; 280 nm).

Yield: 68 mg.

MS(FAB) for C<sub>47</sub>H<sub>62</sub>N<sub>7</sub>O<sub>15</sub>:

Calcd: 964.4304 (M+H);

Found: 964.4348.

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#### Example 5

Preparation of the compound of formula I where R', R" and R" are each methyl, Rx1, Rx2, Ry1, Ry2, Ry3, Ry4, and R<sup>0</sup> are each hydroxy and

 $R^2$  is  $-CH_2$ — $C \equiv C$ —

The desired compound was prepared substantially in accordance with the procedure detailed in Example 4A, using 375.3 mg (0.495 mmol) of the (A-30912A) nucleus (compound of formula IB where R', R" and R" are each methyl,  $R^{x1}$ ,  $R^{x2}$ ,  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$ ,  $R^{y4}$  and  $R^{0}$  are each hydroxy) and 158.3 mg (0.767 mmol) of the compound of Preparation 4 in 10 ml of ethanol.

Yield: 28 mg.

15 MS(FAB) for  $C_{49}H_{60}N_{7}O_{14}$ :

Calcd: 970.4198 (M+H-H<sub>2</sub>O);

Found: 970.4222.

Example 6

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Preparation of the compound of formula I where R'. R" and R" are each methyl, Rx1, Rx2, Rv1, Rv2, Rv3, Rv4, and R0 are each hydroxy and

$$R^2$$
 is  $-CH_2$ — $C\equiv C$ — $C$ 

25 The desired compound was prepared substantially in accordance with the procedure detailed in Example 4A, using 577.8 mg (0.649 mmol) of the (A-30912A) nucleus (compound of formula IB where R', R" and R" are each methyl, R\*1, R\*2, RY1, RY2, RY3, RY4 and R0 is hydroxy) and 164.6 mg (0.583 mmol) of the compound of Preparation 5B in 10 ml of ethanol.

Yield: 51 mg.

MS(FAB) for C<sub>55</sub>H<sub>64</sub>N<sub>7</sub>O<sub>14</sub>:

Calcd: 1046.4511 (M-H<sub>2</sub>O);

35 Found: 1046.4530.

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The compounds of formula I exhibit antifungal and antiparasitic activity. For example, the compounds of formula I inhibit the growth of various infectious fungi including Candida spp. such as C. albicans,

C. parapsilosis, C. krusei, C. glabrata, or C. tropicalis, C. lusitaniae; Torulopus spp. such as T. glabrata;
Aspergillus spp. such as A. fumigatus; Histoplasma spp. such as H. capsulatum; Cryptococcus spp. such as C. neoformans; Blastomyces spp. such as B. dermatitidis; Fusarium spp., Trichophyton spp., Pseudallescheria boydii, Coccidioides immitis, Sporothrix schenckii and the like.

Antifungal activity of a test compound is determined in vitro by obtaining the minimum inhibitory concentration (MIC) of the compound using a standard agar dilution test or a disc-diffusion test. The compound is then tested in vivo (in mice) to determine the effective dose of the test compound for controlling a systemic fungal infection.

Accordingly, the following compounds were tested for antifungal activity against <u>C. albicans</u>.

Table 5

Minimal inhibitory concentration against C. albicans

	•	
25	Example No.	MIC (µq/ml)
	1	0.039
·	2	0.005
	3	5.0
	4	0.312
30	5	20
	6	0.039

In addition, the effective dose of the following compounds for controlling a systemic fungal infection (C. albicans) was tested in vivo (mice).

Table 5 ED<sub>50</sub> (mouse)

	Example No.	ED <sub>50</sub> (mg/kg)
5	1	63
	2	>20
	3	N.T.
	4	>2.5
10	5	>2.5
10	6	>2.5

#### N.T. not tested

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The compounds of the invention also inhibit the growth of certain organisms primarily responsible for opportunistic infections in immunosuppressed individuals. For example the compounds of the invention inhibit the growth of Pneumocystis carinii the causative organism of pneumocystis pneumonia (PCP) in AIDS and other immunocompromised patients. Other protozoans that are inhibited by compounds of formula I include Plasmodium spp., Leishmania spp., Trypanosoma spp., Cryptosporidium spp., Isospora spp., Cyclospora spp., Trichomonas spp., Microsporidiosis spp. and the like.

The compounds of formula I are active in vitro and in vivo and are useful in combating either systemic fungal infections or fungal skin infections. Accordingly, the present invention provides a method of inhibiting fungal activity comprising contacting a compound of formula I, or a pharmaceutically acceptable salt thereof, with a fungus. A preferred method includes inhibiting Candida albicans or Aspergillus fumigatis activity. The present invention further provides a method of treating a fungal infection which comprises administering an effective amount of a compound of formula I, or a pharmaceutically acceptable salt thereof, to a host in need of such treatment. A preferred method includes treating a Candida albicans or Aspergillus fumigatis infection.

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With respect to antifungal activity, the term "effective amount," means an amount of a compound of the present invention which is capable of inhibiting fungal activity. The dose administered will vary depending on such factors as the nature and severity of the infection, the age and general health of the host and the tolerance of the host to the antifungal agent. The particular dose regimen likewise may vary according to such factors and may be given in a single daily dose or in multiple doses during the day. The regimen may last from about 2-3 days to about 2-3 weeks or longer. A typical daily dose (administered in single or divided doses) will contain a dosage level of from about 0.01 mg/kg to about 100 mg/kg of body weight of an active compound of this invention. Preferred daily doses generally will be from about 0.1 mg/kg to about 60 mg/kg and ideally from about 2.5 mg/kg to about 40 mg/kg.

The present invention also provides pharmaceutical formulations useful for administering the antifungal compounds of the invention. Accordingly, the present invention also provides a pharmaceutical formulation comprising one or more pharmaceutically acceptable carriers, diluents or excipients and a compound of claim 1. The active ingredient in such formulations comprises from 0.1% to 99.9% by weight of the formulation, more generally from about 10% to about 30% by weight. By "pharmaceutically acceptable" it is meant that the carrier, diluent or excipient is compatible with the other ingredients of the formulation and not deleterious to the recipient thereof.

A compound of formula I may be administered parenterally, for example using intramuscular, subcutaneous, or intra-peritoneal injection, nasal, or oral means. In addition to these methods of administration, a compound of formula I may be applied topically for skin infections.

For parenteral administration the formulation comprises a compound of formula I and a physiologically

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acceptable diluent such as deionized water, physiological saline, 5% dextrose and other commonly used diluents. The formulation may contain a solubilizing agent such as a polyethylene glycol or polypropylene glycol or other known solubilizing agent. Such formulations may be made up in sterile vials containing the antifungal and excipient in a dry powder or lyophilized powder form. Prior to use, a physiologically acceptable diluent is added and the solution withdrawn via syringe for administration to the patient.

The present pharmaceutical formulations are prepared by known procedures using known and readily available ingredients. In making the compositions of the present invention, the active ingredient will generally be admixed with a carrier, or diluted by a carrier, or enclosed within a carrier which may be in the form of a capsule, sachet, paper or other container. When the carrier serves as a diluent, it may be a solid, semi-solid or liquid material which acts as a vehicle, excipient or medium for the active ingredient. Thus, the compositions can be in the form of tablets, pills, powders, lozenges, sachets, cachets, elixirs, suspensions, emulsions, solutions, syrups, aerosols, (as a solid or in a liquid medium), ointments containing, for example, up to 10% by weight of the active compound, soft and hard gelatin capsules, suppositories, sterile injectable solutions, sterile packaged powders and the like.

For oral administration, the antifungal compound is filled into gelatin capsules or formed into tablets. Such tablets may also contain a binding agent, a dispersant or other suitable excipients suitable for preparing a proper size tablet for the dosage and particular antifungal compound of the formula I. For pediatric or geriatric use the antifungal compound may be formulated into a flavored liquid suspension, solution or emulsion. A preferred oral formulation is linoleic acid, cremophor RH-60 and water and preferably in the amount (by volume) of 8% linoleic acid,

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5% cremophor RH-60, 87% sterile water and a compound of formula I in an amount of from about 2.5 to about 40 mg/ml.

For topical use the antifungal compound may be formulated with a dry powder for application to the skin surface or it may be formulated in a liquid formulation comprising a solubilizing aqueous liquid or non-aqueous liquid, e.g., an alcohol or glycol.

The following formulation examples are illustrative only and are not intended to limit the scope of the invention in any way. The term "active ingredient" means a compound according to formula I or a pharmaceutically acceptable salt thereof.

#### Formulation 1

Hard gelatin capsules are prepared using the following ingredients:

	ļ	Quantity
		<pre>(mg/capsule)</pre>
	Active ingredient	250
20	Starch, dried	200
	Magnesium stearate	<u>10</u>
	Total	460 mg

#### Formulation 2

25 A tablet is prepared using the ingredients below:

		Quantity
		<pre>(mg/capsule)</pre>
	Active ingredient	250
30	Cellulose, microcrystalline	400
	Silicon dioxide, fumed	<b>10</b> .
	Stearic acid	<u>_5</u>
	Total	665 mg

The components are blended and compressed to form tablets each weighing 665 mg.

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#### Formulation 3

An aerosol solution is prepared containing the following components:

		<u>weight</u>
5 .	Active ingredient	0.25
	Methanol	25.75
	Propellant 22	•
	(Chlorodifluoromethane)	<u>74.00</u>
	Total	100.00

units are then fitted to the container.

The active compound is mixed with ethanol and the mixture added to a portion of the propellant 22, cooled to

-30°C and transferred to a filling device. The required amount is then fed to a stainless steel container and diluted with the remainder of the propellant. The valve

#### Formulation 4

Tablets, each containing 60 mg of active

20	ingredient, are made as foll	ows:
	Active ingredient	60 mg
	Starch	45 mg
	Microcrystalline cellulose	35 mg

Polyvinylpyrrolidone

	Polyvinylpyrrolidone	
25	(as 10% solution in water)	4 mg
	Sodium carboxymethyl starch	4.5 mg
	Magnesium stearate	0.5 mg
	Talc	1 mg
	Total	150 mg

The active ingredient, starch and cellulose are passed through a No. 45 mesh U.S. sieve and mixed thoroughly. The aqueous solution containing polyvinyl-pyrrolidone is mixed with the resultant powder, and the mixture then is passed through a No. 14 mesh U.S. sieve.

The granules so produced are dried at 50°C and passed through a No. 18 mesh U.S. sieve. The sodium carboxymethyl starch, magnesium stearate and talc, previously passed

through a No. 60 mesh U.S. sieve, are then added to the granules which, after mixing, are compressed on a tablet machine to yield tablets each weighing 150 mg.

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#### Formulation 5

Capsules, each containing 80 mg of active ingredient, are made as follows:

Active ingredient 80 mg

Starch 59 mg

10 Microcrystalline cellulose 59 mg

Magnesium stearate 2 mg

Total 200 mg

The active ingredient, cellulose, starch and magnesium stearate are blended, passed through a No. 45 mesh U.S. sieve, and filled into hard gelatin capsules in 200 mg quantities.

#### Formulation 6

Suppositories, each containing 225 mg of active ingredient, are made as follows:

Active ingredient

225 mg

Saturated fatty acid glycerides 2.000 mg

Total

2,225 mg

The active ingredient is passed through a No. 60 mesh U.S. sieve and suspended in the saturated fatty acid glycerides previously melted using the minimum heat necessary. The mixture is then poured into a suppository mold of nominal 2 g capacity and allowed to cool.

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#### Formulation 7

Suspensions, each containing 50 mg of active ingredient per 5 ml dose, are made as follows:

Active ingredient 50 mg

Sodium carboxymethyl cellulose 50 mg

Syrup 1.25 ml

Benzoic acid solution 0.10 ml

Flavor q.v.

Color '

q.v.

Purified water to total

5 ml

The active ingredient is passed through a No. 45 mesh U.S. sieve and mixed with the sodium carboxymethyl cellulose and syrup to form a smooth paste. The benzoic acid solution, flavor and color are diluted with a portion of the water and added, with stirring. Sufficient water is then added to produce the required volume.

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#### Formulation 8

An intravenous formulation may be prepared as follows:

Active ingredient

100 mg

Isotonic saline

1,000 ml

The solution of the above ingredients generally is administered intravenously to a subject at a rate of 1 ml per minute.

The present invention further provides a method for treating or preventing the onset of Pneumocystis pneumonia in a host susceptible to Pneumocystis pneumonia which comprises administering an effective amount of a compound of formula I, or a pharmaceutically acceptable salt thereof, to a host in need of such treatment. compounds of formula I can be used prophylactically to prevent the onset of the infection which is caused by the organism Pneumocystis carinii, or alternatively they can be used to treat a host that has been infected with P. carinii. A compound of formula I may be administered parenterally, for example using intramuscular, intravenous or intra-peritoneal injection, orally or by inhaling directly into the airways of the lungs. A preferred mode of administration is inhalation of an aerosol spray formulation of a compound of formula I.

With respect to antiparasitic activity, the term "effective amount," means an amount of a compound of the present invention which is capable of inhibiting parasitic activity. An effective amount of the compound of formula I

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is from about 3 mg/kg of patient body weight to about 100 The amount administered may be in a single daily dose or multiple doses of, for example, two, three or four times daily throughout the treatment regimen. The amount of the individual doses, the route of delivery, the frequency of dosing and the term of therapy will vary according to such factors as the intensity and extent of infection, the age and general health of the patient, the response of the patient to therapy and how well the patient tolerates the drug. It is known that Pneumocystis pneumonia infections in AIDS patients are highly refractory owing to the nature of the infection. For example, in severe, advanced infections the lumenal surface of the air passages becomes clogged with infectious matter and extensive parasite development occurs in lung tissue. A patient with an advanced infection will accordingly require higher doses for longer periods of time. In contrast, immune deficient patients who are not severely infected and who are susceptible to Pneumocystis pneumonia can be treated with lower and less frequent prophylactic doses.

#### **CLAIMS**

## 1. A compound of the formula:

$$R^{y1} \xrightarrow{O} R^{x1} \xrightarrow{H} R^{2}$$

$$R' \xrightarrow{N} O \xrightarrow{HN} O \xrightarrow{NH} O \xrightarrow{N} O \xrightarrow{N} N^{y4}$$

$$R^{x2} \xrightarrow{R^{y3}} R^{y3}$$

$$R^{y1} \xrightarrow{O} R^{x1} \xrightarrow{H} R^{2}$$

$$R^{y2} \xrightarrow{O} R^{y4}$$

$$R^{y1} \xrightarrow{O} R^{y4}$$

$$(1)$$

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wherein:

R' is hydrogen, methyl or -CH<sub>2</sub>C(O)NH<sub>2</sub>;

R" and R" are independently methyl or hydrogen;

R\*1 is hydrogen, hydroxy or -O-R;

R is C<sub>1</sub>-C<sub>6</sub> alkyl, benzyl, -(CH<sub>2</sub>)<sub>2</sub>Si(CH<sub>3</sub>)<sub>3</sub>,

-CH<sub>2</sub>CHOHCH<sub>2</sub>OH, -CH<sub>2</sub>CH=CH<sub>2</sub>, -(CH<sub>2</sub>)<sub>a</sub>COOH, -(CH<sub>2</sub>)<sub>b</sub>NR<sup>z1</sup>R<sup>z2</sup>,

-(CH<sub>2</sub>)<sub>c</sub>POR<sup>z3</sup>R<sup>z4</sup> or -[(CH<sub>2</sub>)<sub>2</sub>O]<sub>d</sub>-(C<sub>1</sub>-C<sub>6</sub>)alkyl;

a, b and c are independently 1, 2, 3, 4, 5 or 6;

R<sup>z1</sup> and R<sup>z2</sup> are independently hydrogen, C<sub>1</sub>-C<sub>6</sub> alkyl,

or R<sup>z1</sup> and R<sup>z2</sup> combine to form -CH<sub>2</sub>(CH<sub>2</sub>)<sub>e</sub>CH<sub>2</sub>-;

R<sup>z3</sup> and R<sup>z4</sup> are independently hydroxy or C<sub>1</sub>-C<sub>6</sub> alkoxy; d is 1 or 2; e is 1, 2 or 3;

 $R^{x2}$ ,  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$  and  $R^{y4}$  are independently hydroxy or hydrogen;

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 $\mathbb{R}^0$  is hydroxy,  $-\mathbb{OP}(\mathbb{O})(\mathbb{OH})_2$  or a group of the formulae:

 $R^1$  is  $C_1$ - $C_6$  alkyl, phenyl, p-halo-phenyl,

p-nitrophenyl, benzyl, p-halo-benzyl or p-nitro-benzyl;

$$R^2$$
 is  $-CH_2 \longrightarrow R^3$ ;

 $\mathbb{R}^3$  is

$$\mathbb{R}^{3a}$$

$$\mathbb{C} = \mathbb{C} - \mathbb{C} - \mathbb{R}^{3c}$$
 , or

$$\mathbb{R}^{3d}$$

 $R^{3a}$ ,  $R^{3b}$ ,  $R^{3c}$  and  $R^{3d}$  are independently hydrogen,  $C_1$ - $C_{12}$  alkyl,  $C_2$ - $C_{12}$  alkynyl,  $C_1$ - $C_{12}$  alkoxy,  $C_1$ - $C_{12}$  alkylthio, halo, or -O- $(CH_2)_m$ - $[O-(CH_2)_n]_p$ -O- $(C_1$ - $C_{12}$  alkyl) or -O- $(CH_2)_q$ -X- $R^4$ ;

m is 2, 3 or 4;

n is 2, 3 or 4;

p is 0 or 1;

15 q is 2, 3 or 4;

X is pyrrolidino, piperidino or piperazino; and  $R^4$  is hydrogen,  $C_1-C_{12}$  alkyl,  $C_3-C_{12}$  cycloalkyl, benzyl or  $C_3-C_{12}$  cycloalkylmethyl;

or a pharmaceutically acceptable salt thereof.

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2. A compound according to claim 1 where: R', R" and R" are each methyl;  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$  and  $R^{y4}$  are each hydroxy;  $R^{x1}$  is hydrogen, hydroxy or -O-R;

R is methyl, benzyl, -CH2CHOHCH2OH, -(CH2)\_bNRz1Rz2 or -(CH2)\_2PORz3Rz4;

b is 2, 3, 4, 5 or 6;

 $R^{z1}$  and  $R^{z2}$  are independently hydrogen or  $C_1$ - $C_4$  alkyl;  $R^{z3}$  and  $R^{z4}$  are independently hydroxy or methoxy;  $R^{x2}$  is hydrogen or hydroxy;

 ${\bf R}^{\bf 0}$  is hydroxy or a group of the formulae:

R1 is methyl;

- p-nitrophenyl, benzyl, p-halo-benzyl or p-nitro-benzyl; or a pharmaceutically acceptable salt thereof.
  - 3. A compound according to claim 2 where:  $\mathbf{R}^{\mathbf{x}\mathbf{1}}$  is hydroxy;
  - R<sup>x2</sup> is hydroxy;

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R<sup>0</sup> is hydroxy;

 ${\bf R}^2$  is a group of the formula:

$$\mathbb{C} = \mathbb{C} - \mathbb{C} + \mathbb{R}^{3c}$$
 , or

$$\mathbb{R}^{3d}$$
 ;

 $R^{3c}$  and  $R^{3d}$  are independently hydrogen,  $C_2$ - $C_{12}$  alkynyl,  $C_1$ - $C_{12}$  alkoxy or -O- $(CH_2)_m$ - $[O-(CH_2)_n]_p$ -O- $(C_1$ - $C_{12}$  alkyl); or a pharmaceutically acceptable salt thereof.

- 4. A compound according to claim 3 where  $R^2$  is  $-CH_2$  or  $-CH_2$   $C\equiv C$
- or a pharmaceutically acceptable salt thereof.
  - 5. A pharmaceutical formulation comprising a compound of formula I, or a pharmaceutically acceptable salt thereof, a claimed in any one of claims 1 to 4,

associated with one or more pharmaceutically acceptable carriers, diluents or excipients therefor.

- 6. A compound of formula I, or a pharmaceutically
  5 acceptable salt thereof, a claimed in any one of claims 1 to 4, for use as a pharmaceutical.
- 7. A process for preparing a compound of formula I, or a pharmaceutically acceptable salt thereof, as claimed in any one of claims 1 to 4, comprising alkylating a compound of formula IB

wherein:

15 R', R", R",  $R^{x1}$ ,  $R^{x2}$ ,  $R^{y1}$ ,  $R^{y2}$ ,  $R^{y3}$ ,  $R^{y4}$  and  $R^0$  are as defined in claim 1.

# INTERNATIONAL SEARCH REPORT

International application No. PCT/US96/07252

A. CLASSIFICATION OF SUBJECT MATTER IPC(6) :CO7K 7/56; A61K 38/12									
According t	:514/9, 11; 530/317 to International Patent Classification (IPC) or to both national cla	esification and IPC							
B. FIELDS SEARCHED									
	ocumentation searched (classification system followed by classifi	cation symbols)							
U.S. : 514/9, 11; 530/317									
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched									
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)									
C. DOCUMENTS CONSIDERED TO BE RELEVANT									
Category*	Category* Citation of document, with indication, where appropriate, of the relevant passages								
X	EP, A, 0,561,639 (ELI LILLY AND COMPAN 1993, page 1 and claim 17.	NY) 22 September	1-7						
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Furthe	er documents are listed in the continuation of Box C.	See patent family annex.							
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(54) Title: RING MODIFIED CYCLIC PEPTIDE ANALOGS

#### (57) Abstract

A method for modifying the cyclic peptide ring system of Echinocandin-type compounds to produce new analogs having antifungal activity is provided. The inventive process comprises opening the cyclic peptide ring, cleaving the terminal ornithine unit, inserting at least one new amino acid or other synthetic unit and closing the ring to produce a new cyclic peptide ring structure. The process allows one to incorporate features such as water-solubility into the cyclic peptide ring nucleus, sites for further modification, increase or decrease the number of amino acid or peptide units within the ring nucleus, and increase or decrease the total number of members within the ring. The invention further provides novel Echinocandin type compounds and their use as antifungal or anti-parasitic agents.

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### RING MODIFIED CYCLIC PEPTIDE ANALOGS

### **CROSS-REFERENCE TO RELATED APPLICATIONS**

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1998.

This is a conversion application of US 60/097,228 filed on August 20,

### **TECHNICAL FIELD**

The present invention relates to the preparation of ring modified cyclic peptide analogs by replacing peptide unit(s) in the cyclic peptide ring nucleus of natural products or semi-synthetic derivatives thereof, in particular, Echinocandin-type compounds, and novel semi-synthetic cyclic peptide compounds produced therefrom.

### **BACKGROUND ART**

Echinocandin B is a natural product with antifungal activity that has been

modified in the past in a variety of ways. For example, simple derivatives have been made including dihydro-and tetrahydro-reduction products and modification of active groups pendant from the ring nucleus. The most common approach has been replacement of the N-acyl side chain. For example, U.S. Patent Nos.

4,293,489; 4,320,052; 5,166,135; and 5,541,160; and EP 359529; 448353; 447186; 462531; and 561639 describe a variety of N-acyl derivatized Echinocandin-type compounds that provide varying degrees of antifungal and antiprotozoal activities.

Other modifications have included acylation of the hydroxyl group of the pendant phenolic group. For example, GB 2,242,194; and EP 448343; 448354; 503960 and 525889 describe the introduction of acyl, phosphono and sulfo radicals having a charged group at neutral pH to impart water solubility.

GB 2,241,956 and EP 448355 describe hydrogen-reduction products of cyclohexapeptide compounds.

A review of the Echinocandin families and their semi-synthetic analogs may be found in Turner, W., et al, *Current Pharmaceutical Design*, 2, 209-224 (1996). The review compares the *in vitro* and *in vivo* activities of the Echinocandin natural products and their semi-synthetic analogs.

Each of the approaches described above are limited to reactions with active

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groups pendant to the cyclic peptide ring nucleus. Some have attempted to build the entire cyclic peptide nucleus synthetically. (See, i.e., U.S. Patent No. 5,696,084; *J. Am. Chem. Soc.*, 108, 6041 (1986); Evans, D.A., et al., *J. Am. Chem. Soc.*, 109, 5151 (1987); *J. Med. Chem.*, 35, 2843 (1992); and Kurokawa, N., et al., *Tetrahedron*, 49, 6195 (1993).) However, this approach is not cost effective and may lead to racemic mixtures. Therefore, there is a need to provide a more flexible and cost effective process for modifying the cyclic hexapeptide nucleus of natural products to broaden the scope of potential antifungal candidates.

Several investigators have disclosed the preferential cleavage of an amide bond in compounds bearing hydroxyl groups in the delta and gamma positions relative to the amide bond to provide asymmetric lactones using acids such as hydrochloric acid and trifluoroacetic acid; however, none have applied the process to the cleavage of a terminal amino acid group of a linear peptide. (See, i.e., K. Tanaka, et al., *Tetrahedron Lett*, **26**(10), 1337 (1985); N. Baba, et al., *Chem Lett* (5), 889 (1989); H. Yoda, et al., *Chem Express*, **4**(8), 515 (1989); and Y. Yamamoto, et al., *J Org Chem*, **56**(3), 112 (1991).)

### BRIEF SUMMARY OF THE INVENTION

The present invention provides a method for modifying the cyclic peptide ring system of Echinocandin-type compounds to produce new analogs having antifungal activity. The inventive process allows one to make changes in the cyclic peptide structure of natural and semi-synthetic products that were previously not possible. For example, one may incorporate features such as water-solubility into the cyclic peptide ring nucleus, sites for further modification, increase or decrease the number of amino acid or peptide units within the ring nucleus, and increase or decrease the total number of members (or atoms) in the ring nucleus.

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The process includes the steps of (i) providing a cyclic peptide compound comprising a peptide unit having a \gamma-hydroxyl group; (ii) opening the ring of the cyclic peptide compound to provide a first linear peptide wherein the peptide unit having a γ-hydroxyl group is the N-terminus peptide unit of the first linear peptide; (iii) cleaving-off the peptide unit having a γ-hydroxyl group to provide a second linear peptide (preferably by adding trifluoroacetic acid or hydrochloric acid to the first linear peptide in an organic solvent); (iv) attaching at least one amino acid, a dipeptide unit or a synthetic unit to the second linear peptide to produce a third linear peptide; (v) cyclizing the third linear peptide to produce a modified cyclic peptide compound having a modified ring nucleus. Alternatively, a second peptide unit may be cleaved-off the second linear peptide produced in step (iii) prior to attaching the amino acid, dipeptide or synthetic unit(s) in step (iv) and subsequent cyclization in step (v). The addition of two or more units in step (iv) may be accomplished in a stepwise fashion (e.g., first one unit is attached then a second unit is attached). Of particular interest is the modification of Echinocandin-type compounds to produce novel cyclic hexapeptide and

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heptapeptide compounds that show inhibition of fungal and parasitic activity. The process also provides a convenient means to produce cyclic peptide compounds having the formulas I and II (including pharmaceutically acceptable salts, esters and hydrates thereof).

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II

wherein

R is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, or heteroaryl group;

R<sup>2</sup> is -H or -CH<sub>3</sub>;

R<sup>3</sup> is -H, -CH<sub>3</sub>, -CH<sub>2</sub>CONH<sub>2</sub> or -CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>;

R<sup>4</sup> is -H or -OH;

 $R^5$  is -OH, -OPO<sub>3</sub>H<sub>2</sub>, or -OSO<sub>3</sub>H;

R<sup>6</sup> is -H or -OSO<sub>3</sub>H;

R<sup>7</sup> is -CH<sub>3</sub> or -H;

(Y) is represented by the following formula

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wherein

A is  $-(CH_2)_a$ - where a=1-4,  $-CHR'-CHR''-(CH_2)_b$ - where R' and R'' are independently -H, -OH, C<sub>6</sub>H<sub>5</sub>O-, -SH,

-NH<sub>2</sub>,  $C_nH_{n+2}NH$ -,  $C_nH_{n+2}O$ -,  $C_nH_{n+2}S$ - or  $-C_nH_{n+2}$  where n = 1-4 and b = 0-1,  $-(CH_2)_c$ -C(O)NH(CH<sub>2</sub>)<sub>d</sub>- where

c = 1-2 and d = 1-2,  $-N=CH-(CH_2)_e$ - where e = 0-2,  $-NR'''(CH_2)_f$ -where R''' is -H,  $-C(O)CH_2NH_2$ ,

-C(O)CH(NH<sub>2</sub>)CH<sub>2</sub>NH<sub>2</sub> or -C<sub>n</sub>H<sub>n+2</sub> where n = 1-4 and f = 1-3, -

 $(CH_2)_g$ -SO<sub>2</sub>- $(CH_2)_h$ - where g = 1-2 and h = 1-2,

where i = 1 or 2, or

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where j is 1 or 2 and Z is an amino group, alkylamino group, or piperidyl amino group;

B is a substituted or unsubstituted C1 to C6 alkyl group (e.g., isopropyl, p-hydroxybenzyl, hydroxymethyl, or α-hydroxyethyl); and W is a hydrogen or C(O)R where R is as defined above.

In another embodiment of the present invention, novel cyclic peptide compounds are provided having the formulas I and II (above) wherein

10 A is -(CH<sub>2</sub>)<sub>a</sub>- where a = 1, 2 or 4, -CHR'-CHR"-(CH<sub>2</sub>)<sub>b</sub>- where R' and R" are independently -H, -OH,  $C_6H_5O$ -, -SH, -NH<sub>2</sub>,  $C_nH_{n+2}NH$ -,  $C_nH_{n+2}O$ -,  $C_nH_{n+2}S$ - or - $C_nH_{n+2}$  where n = 1-4 and b = 0, -(CH<sub>2</sub>)<sub>c</sub>- C(O)NH(CH<sub>2</sub>)<sub>d</sub>- where c = 1-2 and d = 1-2, -N=CH-(CH<sub>2</sub>)<sub>e</sub>- where e = 0-2, -NR"'(CH<sub>2</sub>)<sub>f</sub>- where R"' is -H, -C(O)CH<sub>2</sub>NH<sub>2</sub>, -C(O)CH(NH<sub>2</sub>)CH<sub>2</sub>NH<sub>2</sub> or - $C_nH_{n+2}$  where n = 1-4 and f = 1-3, -(CH<sub>2</sub>)<sub>g</sub>-SO<sub>2</sub>-(CH<sub>2</sub>)<sub>h</sub>- where g = 1-2 and h = 1-2,

where i = 1 or 2, or

where j is 1 or 2 and Z is an amino group, alkylamino group, or piperidyl amino group.

In yet another embodiment of the present invention, a pharmaceutical composition is provided comprising the novel compounds I and II described above (including pharmaceutically acceptable salts, esters and hydrates thereof) in a pharmaceutically inert carrier. Methods for using the novel compounds and pharmaceutical compositions described above for inhibiting fungal growth and parasitic activity are also provided, as well as a method for treating a fungal infection in a human comprising administering to a human in need of such treatment a therapeutically effective amount of the novel antifungal compound described above.

As used herein, the term "Echinocandin-type compounds" refers to compounds having the following general structure including any simple derivatives thereof:

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wherein R is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, or heteroaryl group; R<sup>1</sup> is -H or -OH; R<sup>2</sup> is -H or -CH<sub>3</sub>; R<sup>3</sup> is -H, -CH<sub>3</sub>, -CH<sub>2</sub>CONH<sub>2</sub> or -CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>;

R<sup>4</sup> is -H or -OH; R<sup>5</sup> is -OH, -OPO<sub>3</sub>H<sub>2</sub>, or -OSO<sub>3</sub>H; and R<sup>6</sup> is -H or -OSO<sub>3</sub>H.

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The term "alkyl" refers to a hydrocarbon radical of the general formula  $C_nH_{2n+1}$  containing from 1 to 30 carbon atoms unless otherwise indicated. The alkane radical may be straight, branched, cyclic, or multi-cyclic. The alkane radical may be substituted or unsubstituted. Similarly, the alkyl portion of an alkoxy group or alkanoate have the same definition as above.

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The term "alkenyl" refers to an acyclic hydrocarbon containing at least one carbon-carbon double bond. The alkene radical may be straight, branched, cyclic, or multi-cyclic. The alkene radical may be substituted or unsubstituted.

The term "alkynyl" refers to an acyclic hydrocarbon containing at least one carbon-carbon triple bond. The alkyne radical may be straight, or branched. The alkyne radical may be substituted or unsubstituted.

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The term "aryl" refers to aromatic moieties having single (e.g., phenyl) or fused ring systems (e.g., naphthalene, anthracene, phenanthrene, etc.). The aryl groups may be substituted or unsubstituted. Substituted aryl groups include a chain of aromatic moieties (e.g., biphenyl, terphenyl, phenylnaphthalyl, etc.)

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The term "heteroaryl" refers to aromatic moieties containing at least one heteratom within the aromatic ring system (e.g., pyrrole, pyridine, indole, thiophene, furan, benzofuran, imidazole, pyrimidine, purine, benzimidazole, quinoline, etc.). The aromatic moiety may consist of a single or fused ring system. The heteroaryl groups may be substituted or unsubstituted.

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Within the field of organic chemistry and particularly within the field of organic biochemistry, it is widely understood that significant substitution of

compounds is tolerated or even useful. In the present invention, for example, the term alkyl group allows for substituents which is a classic alkyl, such as methyl, ethyl, propyl, hexyl, isooctyl, dodecyl, stearyl, etc. The term group specifically envisions and allows for substitutions on alkyls which are common in the art, such as hydroxy, halogen, alkoxy, carbonyl, keto, ester, carbamato, etc., as well as including the unsubstituted alkyl moiety. However, it is generally understood by those skilled in the art that the substituents should be selected so as to not adversely affect the pharmacological characteristics of the compound or adversely interfere with the use of the medicament. Suitable substituents for any of the groups defined above include alkyl, alkenyl, alkynyl, aryl, halo, hydroxy, alkoxy, aryloxy, mercapto, alkylthio, arylthio, mono- and di-alkyl amino, quaternary ammonium salts, aminoalkoxy, hydroxyalkylamino, aminoalkylthio, carbamyl, carbonyl, carboxy, glycolyl, glycyl, hydrazino, guanyl, and combinations thereof.

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### DETAILED DESCRIPTION OF THE INVENTION

The synthetic scheme outlined below illustrates the general procedures for modifying the cyclic peptide ring system of Echinocandin-type compounds while maintaining chirality. The cyclic peptide ring of any Echinocandin-type natural product or semi-synthetic derivative can be opened and the terminal ornithine peptide unit cleaved so long as the γ-hydroxyl group of the ornithine peptide unit is present and not blocked. The term "natural product" refers to those secondary metabolites, usually of relatively complex structure, which are of more restricted distribution and more characteristic of a specific source in nature. Suitable natural product starting materials belonging to the Echinocandin cyclic peptide family include Echinocandin B, Echinocandin C, Aculeacin Aγ, Mulundocandin,

Sporiofungin A, Pneumocandin  $A_0$ , WF11899A, and Pneumocandin  $B_0$ . For illustrative purposes, the following synthetic scheme starts with Cilofungin.

As shown above, the cyclic hexapeptide ring (1) is first opened using base catalysis and then reduced with sodium borohydride to give the linear hexapeptide (2). Upon treatment with triethyl silane in trifluoroacetic acid (TFA), the benzylic hydroxyl is removed and the ornithine unit is cleaved to give the linear pentapeptide (3). The linear pentapeptide (3) can now be protected and the primary amide activated to provide an intermediate (4) which can be recyclized with a new amino acid unit or other synthetic unit to produce a new cyclic compound (5). Cyclic compound (5) can be further modified by deprotecting and acylating the pendant amino group (if present) to provide modified cyclic compound (6) having an N-acyl side chain. Those skilled in the art will appreciate that the N-acyl side chain encompasses a variety of side chain moieties known in the art. Suitable side chain moieties include substituted and unsubstituted alkyl groups, alkenyl groups, alkynyl groups, aryl groups, heteroaryl groups and combinations thereof. Preferably, the side chain contains both a linearly rigid section and a flexible alkyl section to maximize antifungal potency. In addition, further modifications can be made on any new functionality introduced by the incorporation of the new amino acid, peptide or synthetic unit(s) containing such new functionality.

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Alternatively, another peptide unit can be cleaved from intermediate (3) to provide a tetrapeptide (7) which can be recyclized with a new amino acid unit, dipeptide unit, or other synthetic unit to produce a new cyclic compound (8). Like cyclic compound (5), compound (8) can also be further modified by deprotecting and acylating the pendant amino group (if present) to attach an N-acyl side chain or modification of any new functionality introduced through the incorporation of the new amino acid, peptide or synthetic units.

As illustrated in the synthetic scheme above, the ring nucleus is selectively opened at the C-terminus L-proline, N-terminus R-ornithine linkage using standard base catalysis well known to those skilled in the art. Once the cyclic hexapeptide is open, then the terminal ornithine amino acid may be cleaved and a new amino acid (or other synthetic unit) attached using standard peptide formation processes (or condensation processes) well known to those skilled in the art. The terminal ornithine unit is cleaved with trifluoroacetic acid or hydrochloric acid in an organic solvent such as methylene chloride, toluene, or dioxane. The preferred reaction condition is trifluoroacetic acid in methylene chloride.

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Any amino acid or peptide unit may be attached to the linear peptide.

Theoretically, it is also possible to condense other synthetic units onto the peptide that are capable of cyclization. For example, a sulfonamide linkage may be formed between a terminal amino group on the linear peptide and a sulfonyl group on a synthetic unit. Any number of other linkages may also be envisioned; however, the pharmaceutical activity of such compounds are currently unknown.

The insertion of a new amino acid, dipeptide unit or other synthetic unit allows one to change the size of the cyclopeptide ring. The number of atoms in the ring system may be increased or decreased from the original 21 membered Echinocandin ring structure depending upon the particular compound(s) inserted into the ring. Theoretically, the ring size is limited only by the configuration of the linear peptide. If the linear peptide is too short or too long, the ends cannot come into close enough proximity to react and the ends may polymerize with another linear peptide rather than close to form a ring. The optimum configuration of the linear peptide for ring closure will vary depending upon the particular amino acids that make-up the peptide structure. For Echinocandin-type

compounds, preferably, the final ring structure contains between 19 to 22 members, more preferably, the final ring structure is a 21- or 22-membered ring, most preferably the final ring structure is a 21-membered ring.

It is well-known that the acyl side chain pendant from the Echinocandin ring structure plays an important role in the activity of both the natural products and semi-synthetic Echinocandin type materials. Consequently, any amino acid or synthetic unit may be used for insertion into the ring so long as the final cyclized product contains at least one amino group capable of acylation.

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When the inserted compound contains more than one unit, the units may be attached one at a time to the linear penta- or tetra-peptide or the individual units can be combined and then added to the linear penta- or tetra-peptide as a block unit. Preferably, the units are added as a block unit to minimize racemization.

Acylation of the amino group may be accomplished in a variety of ways well known to those skilled in the art. For example, the amino group may be acylated by reaction with an appropriately substituted acyl halide, preferably in the presence of an acid scavenger such as a tertiary amine (e.g., triethylamine). The reaction is typically carried out at a temperature between about -20°C to 25°C. Suitable reaction solvents include polar aprotic solvents, such as dioxane or dimethylformamide. Solvent choice is not critical so long as the solvent employed is inert to the ongoing reaction and the reactants are sufficiently solubilized to effect the desired reaction.

The amino group may also be acylated by reaction with an appropriately substituted carboxylic acid, in the presence of a coupling agent. Suitable coupling agents include dicyclohexylcarbodiimide (DCC), N,N'-carbonyldiimidazole, bis(2-oxo-3-oxazolidinyl)phosphinic chloride (BOP-Cl), N-ethoxycarbonyl-2-

ethoxy-1,2-dihydroquinoline (EEDQ), benzotriazole-1-yloxytripyrrolidinophosphonium hexafluorophosphate (PyBOP) and the like.

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Alternately, the amino group may be acylated with an activated ester of a carboxylic acid such as p-nitrophenyl, 2,4,5-trichlorophenyl, hydroxybenzotriazole hydrate (HOBT H<sub>2</sub>O), pentafluorophenol, and N-hydroxysuccinimide carboxylate esters. Preferred acylating moieties are the 2,4,5-trichlorophenyl and HOBT carboxylate esters. The reaction is typically ran 1 to 65 hours at a temperature from about 0°C to 30°C in an aprotic solvent. The reaction is generally complete after about 24 to 48 hours when carried out at a temperature between about 15°C to 30°C. Suitable solvents include tetrahydrofuran and dimethylformamide or mixtures thereof. The amino group is generally present in equimolar proportions relative to the activated ester or with a slight excess of the amino group.

The compounds of the present invention may be isolated and used per se or in the form of its pharmaceutically acceptable salt or hydrate. The term "pharmaceutically acceptable salt" refers to non-toxic acid addition salts derived from inorganic and organic acids. Suitable salt derivatives include halides, thiocyanates, sulfates, bisulfates, bisulfates, arylsulfonates, alkylsulfates, phosphonates, monohydrogenphosphates, dihydrogenphosphates,

 metaphosphates, pyrophosphonates, alkanoates, cycloalkylalkanoates, arylalkonates, adipates, alginates, aspartates, benzoates, fumarates, glucoheptanoates, glycerophosphates, lactates, maleates, nicotinates, oxalates, palmitates, pectinates, picrates, pivalates, succinates, tartarates, citrates, camphorates, camphorsulfonates, digluconates, trifluoroacetates, and the like.

The ring-modified compounds may be used in a variety of pharmaceutical formulations. A typical formulation comprises the ring-modified compound (or

its pharmaceutically acceptable salt, ester or hydrate) in combination with a pharmaceutically acceptable carrier, diluent or excipient. The active ingredient is typically formulated into pharmaceutical dosage forms to provide an easily controllable dosage of the drug and to give the patient an elegant and easily handleable product. Formulations may comprise from 0.1% to 99.9% by weight of active ingredient, more generally from about 10% to about 30% by weight.

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As used herein, the term "unit dose" or "unit dosage" refers to physically discrete units that contain a predetermined quantity of active ingredient calculated to produce a desired therapeutic effect. When a unit dose is administered orally or parenterally, it is typically provided in the form of a tablet, capsule, pill, powder packet, topical composition, suppository, wafer, measured units in ampoules or in multidose containers, etc. Alternatively, a unit dose may be administered in the form of a dry or liquid aerosol which may be inhaled or sprayed.

The dosage to be administered may vary depending upon the physical characteristics of the patient, the severity of the patient's symptoms, and the means used to administer the drug. The specific dose for a given patient is usually set by the judgment of the attending physician.

Suitable carriers, diluents and excipients are well known to those skilled in the art and include materials such as carbohydrates, waxes, water soluble and/or swellable polymers, hydrophilic or hydrophobic materials, gelatin, oils, solvents, water, and the like. The particular carrier, diluent or excipient used will depend upon the means and purpose for which the active ingredient is being applied. The formulations may also include wetting agents, lubricating agents, emulsifiers, suspending agents, preservatives, sweeteners, stabilizers, perfuming agents, flavoring agents and combinations thereof.

A pharmaceutical composition may be administered using a variety of methods. Suitable methods include topical (e.g., ointments or sprays), oral, injection and inhalation. The particular treatment method used will depend upon the type of infection being addressed.

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Echinocandin-type compounds have been shown to exhibit antifungal and antiparasitic activity such as growth inhibition of various infectious fungi including Candida spp. (i.e., C. Albicans, C. Parapsilosis, C. Krusei, C. Glabrata, C. Tropicalis, or C. Lusitaniaw); Torulopus spp. (i.e., T. Glabrata); Aspergillus spp. (i.e., A. Fumigatus); Histoplasma spp. (i.e., H. Capsulatum); Cryptococcus spp. (i.e., C. Neoformans); Blastomyces spp. (i.e., B. Dermatitidis); Fusarium spp.; Trichophyton spp., Pseudallescheria boydii, Coccidioides immits, Sporothrix schenckii, etc.

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Compounds of this type also inhibit the growth of certain organisms primarily responsible for opportunistic infections in immunosuppressed individuals, such as growth inhibition of *Pneumocystis carinii* (the causative organism of pneumocystis pneumonia (PCP) in AIDS and other immunocompromised patients. Other protozoans that are inhibited by Echinocandin-type compounds include Plasmodium spp., Leishmania spp., Trypanosoma spp., Cryptosporidium spp., Isospora spp., Cyclospora spp., Trichomnas spp., Microsporidiosis spp., etc.

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The compounds of the present invention are useful in combating either systemic fungal infections or fungal skin infections. Accordingly, a method is provided for inhibiting fungal activity comprising contacting a compound of formula I or II (or a pharmaceutically acceptable salt, ester or hydrate thereof) with a fungus. A preferred method includes inhibiting *Candida albicans* or *Aspergillus funigatis* activity. The term "contacting" includes a union or junction, or apparent touching or mutual tangency of a compound of the invention

with a parasite or fungus. The term does not imply any further limitations to the process, such as by mechanism of inhibition. The methods are defined to encompass the inhibition of parasitic and fungal activity by the action of the compounds and their inherent antiparasitic and antifungal properties.

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A method for treating a fungal infection which comprises administering an effective amount of a compound of formula I or II (or a pharmaceutically acceptable salt, ester or hydrate thereof) to a host in need of such treatment is also provided. A preferred method includes treating a *Candida albicans* or *Aspergillus fumigatis* infection. The term "effective amount" refers to an amount of active compound which is capable of inhibiting fungal activity. The dose administered will vary depending on such factors as the nature and severity of the infection, the age and general health of the host and the tolerance of the host to the antifungal agent. The particular dose regimen likewise may vary according to these factors. The medicament may be given in a single daily dose or in multiple doses during the day. The regimen may last from about 2-3 days to about 2-3 weeks or longer. A typical daily dose (administered in single or divided doses) contains a dosage level between about 0.01 mg/kg to 100 mg/kg of body weight of an active compound. Preferred daily doses are generally between about 0.1 mg/kg to 60 mg/kg and more preferably between about 2.5 mg/kg to 40 mg/kg.

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Although the compounds described herein may be used for inhibiting fungal and parasitic activity in a variety of circumstances (e.g., humans, animals, agriculture, etc.), preferably, the methods of use are limited to the treatment of humans to reduce the potential for developing resistance to the pharmaceutical.

### **EXAMPLES**

Unless indicated otherwise, all chemicals can be acquired from commercial suppliers such as Aldrich Chemical (Milwaukee, WI), Sigma, and other commercial sources well-known to those skilled in the art. The following acronyms are representative of the corresponding functional groups or compounds:

BOC = t-butoxycarbonyl,  $(CH_3)_3C-O-C(O)$ -

 $CBZ = benzyloxycarbonyl, C_6H_5CH_2-O-C(O)-$ 

o-Cl-CBZ = ortho-chlorobenzyloxycarbonyl

FMOC = fluorenylmethyloxycarbonyl

TBDMS = t-butyldimethylsilyl

TFA = trifluoroacetic acid

AcN = acetonitrile

DMF = dimethylformamide

THF = tetrahydrofuran

TDM = 4,4'-tetramethyl-diamino-diphenylmethane

CAM = ceric ammonium molybdate

The following set of examples illustrate the general reaction conditions for cleaving and inserting new unit(s) into a cyclohexapeptide nucleus.

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### **Preparation of Key Intermediates**

Ring opening and reduction of cilofungin (1) to give intermediate I-2.

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<u>I-2</u>

To a stirred solution of Cilofungin (1) (100 g; 96 mmol) in 350 ml of 55% acetonitrile/45% water was added 1 N sodium hydroxide solution (40 ml). The reaction was monitored by high pressure liquid chromatography (C-18 column, 50% AcN/water, 230 nm). After 1 hour, the reaction mixture contained >90% of the intermediate aldehyde. Next, sodium borohydride (1.8 g; 48 mmol) was added and the stirring continued for 20 min. HPLC showed complete conversion to the final alcohol product. The reaction was quenched by adding acetic acid dropwise until the evolution of gas was complete. Most of the acetonitrile was removed by rotary evaporation followed by lyophilization to remove the remainder to give 98.1 g of a mixture of the solid product 1-2 and inorganic salts. (93% pure by HPLC)

## Peptide cleavage and deoxygenation of I-2 to give pentapeptide (I-3)

I-3

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The unpurified mixture of Compound I-2 (98.1 g) described above was dissolved in trifluoroacetic acid (300 ml) and dichloromethane (100 ml). The mixture was cooled in an ice bath. Triethylsilane (32 ml; 0.2 mol) was added and the reaction was stirred at 0°C for 1 hour. The ice bath was removed and the reaction was left at ambient temperature for 18 hrs. The solvent was removed in vacuo and the residue redissolved in methanol for HPLC purification. The residue was purified by passage through a C-18 column with 50% acetonitrile/water; 0.1% TFA to remove more lipophilic byproducts. The polar peaks were purified with 10% acetonitrile/water; 0.1% TFA. Lyophilization gave 57.8 g (98% yield) of pure pentapeptide trifluoroacetic acid salt (I-3). FAB MS = 653.3 (M+1)

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### Preparation of CBZ pentapeptide (1-4)

<u>I-4</u>

To an ice bath cooled solution of <u>I-3</u> (50.3 g, 66 mmol) in water (200 ml) and tetrahydrofuran (100 ml) was added excess solid sodium bicarbonate until no additional foaming occurred and pH >8. Carbobenzyloxy chloride (10 ml, 70 mmol) was added and the reaction was monitored by HPLC (25% AcN/water, 0.1% TFA, 230 nm). The pH was monitored and occasionally more sodium bicarbonate was added to keep the solution basic. After 1 hour, the reaction was complete and the solvent was removed *in vacuo*. The residue was slurried in methanol, the solid inorganics removed by filtration, and the solution was passed through a preparative HPLC (25% methanol/water). Removal of solvents gave 25.2 g (49% yield) of <u>I-4</u> as a white foam. FAB MS = 787.38 (M+1)

## Preparation of Silyl CBZ pentapeptide (I-5)

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<u>I-5</u>

Compound <u>I-4</u> (25.2 g, 32 mmol), imidazole (20.6 g, 303 mmol), and t-butyldimethylsilyl chloride 45.6 g, 303 mmol) in dimethylformamide (250 ml) were mixed while following the reaction by TLC (25% ethyl acetate/hexane). After 6 hours, the solvent was removed *in vacuo* and the residue was slurried and sonicated in ether. The ether solution was washed with 1N HCl, dried over MgSO<sub>4</sub> and reduced *in vacuo* to give a foam. The crude product was purified by flash chromatography (600 g silica, 25% ethyl acetate/hexane) to give 32.9 g (70% yield) of a white foam. NMR data was consistent with the structure <u>I-5</u>. FAB MS = 1472.9 (M)

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## Preparation of DiBOC CBZ silyl pentapeptide (I-6)

<u>I-6</u>

Compound I-5 (32.9 g, 22.3 mmol) was dissolved in acetonitrile (250 ml) and tetrahydrofuran (50 ml). Di-t-butyl dicarbonate (16.1 g, 73.6 mmol) and dimethylaminopyridine (299 mg, 2.2 mmol) were added with stirring. The reaction was followed by TLC (20% ethyl acetate/hexane) and an additional 3 g of di-t-butyl dicarbonate was added after 2 hrs. After an additional 2.5 hours, several ml of acetic acid were added to quench the dimethylaminopyridine. The solvent was removed *in vacuo* keeping the temperature less than  $40^{\circ}$ C. The residue was chromatographed (500 g silica, 15% ethyl acetate/hexane) to give 33.6 g (89% yield) of a white foam. NMR data was consistent with the structure I-6. FAB MS = 1672.0 (M)

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## Preparation of the Intermediate linear tetrapeptide (I-7):

I-7

To a solution of Pentapeptide Compound <u>I-3</u> (5.75 g, 7.50 mmol) in anhydrous DMF (200 ml) was added NaHCO<sub>3</sub> (690 mg, 8.25 mmol) and phenyl isothiocyanate (0.99 ml, 8.25 mmol), and the reaction stirred at room temperature for 36 hours. The solids were removed by filtration and the filtrate was concentrated *in vacuo*. The resulting oil was dissolved in TFA (70 ml) and stirred at room temperature for 1 hour, followed by removal of the solvent in vacuo. The resulting solids were treated with water (150 ml), sonicated, and the insoluble materials were removed by filtration. Reverse phase HPLC of the filtrate (eluting with 4%AcN/0.1% TFA/H<sub>2</sub>O) followed by lyophilization gave 3.20 g of a fluffy white solid, 64% yield. The <sup>1</sup>H NMR (300 MHz)spectrum was consistent with the structure <u>I-7</u>. FAB MS (M<sup>+</sup> of free base)= 552.

Example 1 illustrates the general reaction conditions for converting pentapeptide intermediate (I-6) into a cyclic hexapeptide analog of an Echinocandin-type compound.

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## Example 1

Preparation of DiBOC silyl o-Cl-CBZ hexapeptide (E1-1)

## E1-1

A solution of N- $\alpha$ -BOC-N- $\gamma$ -(2-chloro CBZ)-L-ornithine (480 mg, 1.2 10 mmol), N-hydroxysuccinimide (138 mg, 1.2 mmol), and dicyclohexylcarbodiimide (247 mg, 1.2 mmol) in 4 ml of tetrahydrofuran was stirred overnight to form the active ester. A solution of 1-6 (1.0 g, 0.598 mmol) in ethanol (5 ml) was added to a slurry of 10% Pd/C (250 mg) in 5 ml of ethanol followed by 10 ml of glacial acetic acid. The mixture was put under a balloon of H<sub>2</sub> and after 1 hour the starting material was gone. The catalyst was removed by filtration and the solution was carefully reduced under high vacuum keeping the

temperature under 40°C. The resulting oil was dissolved in ether and the previously prepared tetrahydrofuran solution of active ester was added followed by excess triethylamine until the solution was basic to pH paper. After stirring for 2 hours, the solution was extracted with saturated NaHCO<sub>3</sub> solution followed by dilute HCl solution and then another portion of saturated NaHCO<sub>3</sub> solution. The organic layer was dried over MgSO<sub>4</sub> and reduced *in vacuo* to give 0.85 g of the crude product. Purification by flash chromatography (25% ethyl acetate/hexane) gave 0.53 g of coupled product E1-1 (47% yield). NMR data was consistent with the structure

Cyclization of E1-1 to BOC silyl cyclohexapeptide (E1-2)

E1-1. FAB MS = 1922.2 (M+1)

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E1-2

An ethanol/acetic acid solution (10 ml of each) of  $\underline{E1-1}$  (0.53 g, 0.27 mmol) with 10% Pd/C (200 mg) was placed under a balloon of hydrogen. After 2 hours, TLC (30% ethyl acetate/hexane) indicated a complete reaction. The

catalyst was removed by filtration and the solvent reduced *in vacuo* at  $40^{\circ}$ C until the residue was a thick oil. The residue was dissolved in ethyl ether (150 ml) and excess triethylamine was added until the solution was basic to pH paper (~2 ml). After 18 hours, TLC indicated a single product spot. The solvent was removed *in vacuo* and the residue purified over a flash column to provide 343 mg of a white solid (81% yield). NMR data was consistent with the structure <u>E1-2</u>. FAB MS = 1536.0 (M+1)

### . Removal of protecting groups and coupling of the side chain to give E1-3

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#### E1-3

Compound E1-2 (510 mg, 0.332 mmol) was dissolved in 5 ml trifluoroacetic acid at 0°C. After 0.5 hour, water (0.5 ml) was added and the mixture stirred for 0.5 hour longer. The solvent was removed *in vacuo* and the residue was dissolved in 1N HCl (2 ml) and tetrahydrofuran (2 ml). The solution was refrigerated for 48 hours after which HPLC analysis (15% AcN/water, 230 nm) showed a single product peak. The solvent was removed under high vacuum giving a foam residue which was dissolved in dimethylformamide (8 ml). The

terphenyl hydroxybenzotriazole active ester (191 mg, 0.4 mmol) and triethylamine (0.2 ml, 1.4 mmol) were added to the solution. After 4 hours, HPLC (60% AcN/water, 230 nm) showed complete conversion to a new product peak. The solvent was removed under high vacuum and purified by preparative HPLC using the analytical conditions. Solvent removal from the pure fractions gave 238 mg (66% yield) of a white solid. NMR data was consistent with the structure E1-3. FAB MS calculated for  $C_{58}H_{74}N_7O_{14}$  1092.5294; found 1092.5301 (M).

The following examples provide further illustrations of converting key intermediate (I-6) into a cyclic hexapeptide analog. In a similar manner I-6 was converted to each of the following cyclic peptides:

Coupling with N $\alpha$ -BOC-N $\delta$ -CBZ-D-omithine and subsequent cyclization gave 63.9 mg of E1-4 (21-membered ring). FAB MS calculated for C<sub>58</sub>H<sub>74</sub>N<sub>7</sub>O<sub>14</sub> 1092.5294; found 1092.5280.

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E1-4

Coupling with N $\alpha$ -BOC-N $\epsilon$ -CBZ-L-lysine and subsequent cyclization gave 44.1 mg of E1-5 (22-membered ring). FAB MS calculated for C<sub>59</sub>H<sub>76</sub>N<sub>7</sub>O<sub>14</sub> 1106.5450; found 1106.5464.

E1-5

Coupling with N $\alpha$ -FMOC-N $\delta$ -CBZ-L-2,4-diaminobutyric acid and subsequent cyclization gave 65.0 mg of E1-6 (20-membered ring). FAB MS calculated for  $C_{57}H_{72}N_7O_{14} = 1078.5137$ ; found 1078.5128.

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<u>E1-6</u>

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Coupling with N $\alpha$ -BOC-N $\beta$ -CBZ-L-2,3-diaminopropionic acid and subsequent cyclization gave 25.5 mg of E1-7 (19-membered ring). FAB MS calculated for C<sub>56</sub>H<sub>70</sub>N<sub>7</sub>O<sub>14</sub> 1064.4981; found 1064.4994.

E1-7

Table 1 summarizes the activity data for compounds E1-3 through E1-7 in

comparison with the following comparative semi-synthetic Echinocandin compound C1 which has proven in vitro and in vivo antifungal activity. In a murine model of organ recovery, Compound C1 significantly reduced the number of A. Fumigatus recovered from the kidneys and was as effective as amphotericin B on a mg/kg basis when both were administered intraperitoneally. In a Pneumocystis carinii model, Compound C1 reduced the number of cysts in the lungs of heavily infected, immunosuppressed rats by more than 99% when administered orally at 5 mg/kg once daily for 4 days. Prophylactic oral administration of 1 mg/kg twice daily for 4 weeks resulted in >90% reduction in

all life cycle forms. (see Turner, W.W. and M.J. Rodriguez, Current

Pharmaceutical Design, 1996, 2, p214.)

<u>C1</u>

Antifungal activity of the comparative and test compounds were determined in vitro by obtaining the minimum inhibitory concentration (MIC) of the compound using a standard agar dilution test or a disc-diffusion test.

Table 1

I able 1						
Example No.	Minimal Inhibitory Concentration MIC (μg/ml)					
	C. albicans	C. parapsilosis	A. fumigatus	Histoplasma capsulatum		
Comparative C1 (21-membered ring)	0.01	0.156	0.02	0.01		
E1-3 (21-membered ring)	0.005	0.156	0.078	0.156		
E1-4 (21-membered ring)	1.25	>20	20.0	5.0		
E1-5 (22-membered ring)	0.02	>20	2.5	0.156		
E1-6 (20-membered ring)	0.039	>20	20	0.078		
E1-7 (19-membered ring)	0.312	>20	0.625	0.625		

Example 2 illustrates the conversion of intermediate <u>I-6</u> into an azacyclic hexapeptide analog of an Echinocandin-type compound.

### Example 2

Preparation of the N-BOC, Benzyl, Aldehyde Derivative of
L-Homoserine (E2-1)

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E2-1

The following procedure described in Baldwin & Flinn, Tetrahedron Lett., 26(31), 3605, (1987) was used to prepare E2-1. A suspension of L-homoserine (5 g, 42 mmol) in 20 ml of water was treated with solid sodium bicarbonate (3.5 g, 42 mmol). The mixture was stirred at room temperature for approx. 10 minutes. A solution of di-t-butyl dicarbonate (BOC anhydride) (13.75 g, 63 mmol) in 20 ml of p-dioxane was added to the mixture and then stirred vigorously at room temperature for approx. 60 hours. The resulting homogenous solution was reduced in vacuo to yield a colorless oil. A solution of benzyl bromide (10.8 g, 63 mmol) in 50 ml of dimethyl formamide was added to the residue. Another 1.8 g (0.5 eq more) of solid sodium bicarbonate was added to the reaction and the mixture was allowed to stir at room temperature overnight. The reaction was monitored by thin layer chromatography (1:1 chloroform/methanol, plus 1 drop of glacial acetic acid, developed using TDM stain). The volatiles were removed in vacuo and ethyl acetate was added to the resulting residue. The organic layer was washed with water (2 times), then brine. The organic layer was dried over sodium sulfate, filtered and concentrated to yield 14.7 g of a light-yellow oil. The residue was dissolved in 50 ml of dimethyl sulfoxide. Triethylamine (12.7 g, 126 mmol) was added and the mixture was cooled in an ice-bath. A suspension of sulfur

trioxide/pyridine complex (20 g, 126 mmol) in 50 ml of dimethylsulfoxide was added with stirring. The ice-bath was removed and the mixture was allowed to warm to room temperature. After approx. 10 minutes, the reaction mixture was poured into 200 ml of ice-water. The aqueous layer was extracted twice with ethyl acetate, the ethyl acetate extract was washed once with 0.1N sodium bisulfate, once with water and then finally with brine. The organic layer was dried over sodium sulfate, filtered and concentrated *in vacuo* to yield a yellow oil. Flash silica gel column purification chromatography (approx. 250 g, 30% ethyl acetate/hexane) yielded 11.1 g (86%) of a light-yellow oil. NMR and elemental analysis (C,H,N) data were consistent with structure <u>E2-1</u>. MS(FD+) = 308 (M+H)

## Preparation of the Dimethyl Acetal of Compound E2-1 (E2-2)

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E2-2

Compound E2-1 (4 g) was dissolved in 50 ml of methanol and then cooled in an ice-bath. A light blanket of HCl gas was introduced over the stirring solution (approx. 2-3 seconds of gas allowed in) and the solution was allowed to continue to stir cold. The reaction was monitored by TLC (30% ethyl acetate/hexane, CAM stain). Additional amounts of HCl gas were added as necessary to drive the reaction to completion. After approx. 2 hours, the reaction appeared to be complete. While still cold, the reaction was quenched by adding solid sodium bicarbonate and keeping the pH slightly on the basic side. The volatiles were

removed *in vacuo*. Ether was added to the residue. The ether layer was washed once with saturated sodium bicarbonate. The aqueous layer was back-extracted with ether. The combined ether extracts were then washed once with brine and dried over sodium sulfate, filtered and concentrated *in vacuo* to yield 4.6 g (100%) of a clear, colorless oil. NMR and elemental analysis (CHN) data were consistent with the structure <u>E2-2</u>.

MS(FD+) = 354 (M+H)

### Debenzylation of Compound E2-2 (E2-3)

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### E2-3

Compound E2-2 (4.2 g, 11.9 mmol) was dissolved in 50 ml of ethyl acetate. The solution was evacuated/purged with nitrogen 3 times, then 1.9 g of 5% palladium on charcoal catalyst was added. The flask was evacuated one more time and then hydrogen was introduced into the flask. The reaction was monitored by TLC (40% ethyl acetate/hexane) and after approx. 2 hours, the reaction was complete. The hydrogen was removed *in vacuo*, the solution was purged with nitrogen, Celite filter aid was added, stirred, filtered through a small bed of Celite on a sintered glass funnel, rinsed with ethyl acetate and the filtrate concentrated *in vacuo* to yield 3.1 g (100%) of a colorless oil. NMR data was consistent with the structure E2-3. MS(FD+) = 264 (M+H) Elemental Analysis (CHN): Theoretical % (C - 50.18; H - 8.04; N - 5.32) Observed % (C - 51.14; H - 7.56; N - 5.65)

Coupling of Compound 2-3 to the Di-BOC-Silyl-CBZ Pentapeptide (I-6) to give (E2-4)

E2-4

Compound <u>I-6</u> was dissolved in 5 ml of ethanol and added to a slurry of 200 mg of 10% Palladium on charcoal in 10 ml of ethanol (all under an atmosphere of nitrogen). Glacial acetic acid (2 ml) was added and then a hydrogen atmosphere was introduced via a balloon.

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Meanwhile, compound <u>E2-3</u> (0.2 g, 0.76 mmol) was dissolved in 2 ml of tetrahydrofuran (Sure Seal, or freshly distilled from lithium aluminum hydride), 96 mg (0.84 mmol) of N-hydroxysuccinimide was added, followed by 172 mg (0.84 mmol) of dicyclohexylcarbodiimide. The reaction mixture was stirred at room temperature. After approx. 10 minutes, a heavy precipitate was observed. Both reactions were allowed to stir at room temperature approx. 2 to 3 hours and monitored by TLC (25% ethyl acetate/hexane, CAM stain).

After completion of the hydrogenation reaction, the mixture was purged with nitrogen, Celite filter aid was added, stirred, and then filtered through a bed of Celite in a sintered glass funnel. The filtrate was concentrated *in vacuo*, not letting the bath temperature rise above 45°C. The residue was dissolved in 8 ml of

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tetrahydrofuran and then approx. 2 ml of triethylamine was added to bring the pH to between 6 to 7. The newly formed active ester from the second reaction was filtered directly into this vessel and enough triethylamine was added to keep the reaction mixture basic (approx. pH 9 - 10). The reaction was stirred at room temperature overnight. The reaction monitored by TLC (25% ethyl acetate/hexane, CAM stain).

The volatiles were removed *in vacuo*, chloroform was added to the residue, the organic layer was washed once with 1N hydrochloric acid, once with saturated sodium bicarbonate solution, once more time with 1N hydrochloric acid, and finally once with brine. The solution was dried over sodium sulfate, filtered, and concentrated *in vacuo* to yield 1 g of E2-4 as a white foam. The material was used without further purification in the subsequent reaction. However, purification can be accomplished using flash silica gel purification chromatography (approx. 100 g of silica, 20% ethyl acetate/hexane). Yield = 395 mg (37%) of a white foam. NMR data was consistent with structure E2-4. MS(FAB) = 1726 (M - t-Butyl)

# Preparation of the Acyl Hydrazone of Compound E2-4 (E2-5)

### E2-5

A flask was charged with compound E2-4 (374 mg, 0.21 mmol) and 8 ml of tetrahydrofuran. Hydrazine hydrate (13.6 mg, 0.27 mmol, 13.6  $\mu$ l) was added and stirred at room temperature. The reaction was monitored by TLC (25% ethyl acetate/hexane, CAM stain). After approx. 15 minutes, the volatiles were removed to yield a white foam. Flash silica gel column chromatography (approx. 25 g, 25%  $\rightarrow$  50% ethyl acetate/hexane) yielded 250 mg (75%) of a white foam. NMR data was consistent with the structure E2-5. MS(FAB) = 1541(M - t-Butyl)

### 10 Cyclization of Compound E2-5 (E2-6)

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E2-6

Stannous chloride (1.5 g) and solid sodium bicarbonate (400 mg) was

suspended in 800 ml of methylene chloride. The mixture was stirred for approx.

15 minutes at room temperature. A solution of compound <u>E2-5</u> (3.2 g) was added in 100 ml of methylene chloride. The container containing compound <u>E2-5</u> was rinsed with another 100 ml of solvent and added to the reaction vessel. The

reaction was monitored by TLC (40% ethyl acetate/hexane, CAM stain). After approx. 3 hours, the residual solids were filtered off and the filtrate concentrated in vacuo to yield 3 g (100%) of  $\underline{E2-6}$  as a pale-yellow solid. Any attempt at purification failed due to instability. MS(FAB) = 1534 (parent ion)

Borohydride Reduction of Compound E2-6 (E2-7)

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E2-7

Compound E2-6 (6 g, 3.9 mmol) was dissolved in 400 ml of tetrahydrofuran, 425 mg (6.76 mmol) of sodium cyanoborohydride was added followed by 1 ml of glacial acetic acid. The mixture was allowed to stir at room temperature while monitoring by TLC (40% ethyl acetate/hexane, CAM stain). After approx. 2 hours, the volatiles were removed *in vacuo*, ethyl acetate was added to the residue, washed twice with water, once with brine, dried over sodium sulfate, filtered, and concentrated *in vacuo* to yield 5.2 g of a white foam. The solid was dissolved in 125 ml of methanol and allowed to stir at room temperature for 6 days. The volatiles were removed *in vacuo* to yield 4.8 g of a white foam. Flash silica gel

column chromatography (approx. 200 g,  $25\% \rightarrow 35\%$  ethyl acetate/hexane) yielded 2.2 g (37%) of a white foam. NMR data was consistent with the structure E2-7. MS(FAB+) = 1537 (M+H)

# CBz Protection of Compound E2-7 (E2-8)

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E2-8

A solution of Compound E2-7 (650 mg, 0.4 mmol) in 15 ml of tetrahydrofuran and solid sodium bicarbonate (67 mg) were added to a reaction vessel fitted with rubber septum and stir bar. The solution was flushed with nitrogen and stirred. The mixture was cooled in an ice-bath and benzyl chloroformate (144 mg, 0.8 mmol, 12 μl) was added via syringe. The mixture was stirred cold for approx. 1 hour. The ice-bath was removed and the mixture was allowed to stir at room temperature for approx. 3 hrs. The reaction was monitored by TLC (20% and 40% ethyl acetate/hexane, CAM stain). The volatiles were removed *in vacuo*. The residue was dissolved in ether, washed twice with water, once with dilute hydrochloric acid quickly, once with saturated

sodium bicarbonate solution, once with brine, dried over sodium sulfate, filtered, and concentrated *in vacuo* to yield 700 mg of a white foam. Flash silica gel column chromatography (35 g of silica, 20% ethyl acetate/hexane) yielded 460 mg (65%) of a white foam. NMR data was consistent with the structure E2-8. MS(FAB) = 1670 (parent ion)

# Deprotection of Compound E2-8 (E2-9)

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E2-9

Compound E2-8 (1.2 g, 0.72 mmol) was dissolved in cold trifluoroacetic acid (10 ml), placed in an ice-bath and stirred for approx. 1.5 hrs. Cold water (5 ml) was added and stirring continued in the ice-bath for approx. 1 hr. The volatiles were removed *in vacuo*, 5 ml of tetrahydrofuran and 5 ml of 1N hydrochloric acid was added and stirred overnight at room temperature.

The volatiles were removed in vacuo, toluene was added to the residue and then evaporated. The procedure was repeated two more times in order to remove the excess trifluoroacetic acid. Ether was added to the residue and then sonicated to yield a white solid. The ether was filtered off and the residue rinsed several

times with ether. The residue was dried under high vacuum to yield 660 mg (100%) of a white solid. Analysis by RP-HPLC (C-18 Bondapak, 70:20:10 AcN/water/1% TFA, 230 nm) shows material to be 97% pure. NMR data was consistent with the structure  $\underline{\text{E2-9}}$ . MS(FAB) = 885.5 (M+H)

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# Preparation of compound(E2-10(b))

E2-10(b)

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Compound E2-9 (600 mg, 0.68 mmol) was dissolved in 10 ml of dimethylformamide and enough diisopropyl ethylamine added to make the solution basic to pH paper (approx. 0.5 ml). The hydroxybenzotriazole active ester of the terphenyl side chain (388 mg, 0.81 mmol) was added to the reaction vessel and allowed to stir at room temperature overnight. The reaction was monitored by RP-HPLC (60:40 AcN/water, 230 nm).

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The solvent was removed *in vacuo*, and a 1:1 mixture of methanol/acetonitrile was added to the resulting residue, followed by stirring and then filtration. The resulting white solid was suspended in ether, stirred, filtered, and the process repeated. The same procedure was performed using methylene chloride, then finally one time more with ether. The residue was dried under vacuum to yield

735 mg (88%) of a white solid - Compound  $\underline{E2-10(a)}$ . MS(FAB)= 1227.6 (parent ion)

The white solid (632 mg, 0.5 mmol) was suspended in glacial acetic acid (100 ml) and subjected to catalytic hydrogenation under a balloon of hydrogen, overnight (100 mg of 10% palladium on charcoal). The reaction was monitored by reverse-phase HPLC (C-18 Bondapak, 60:30:10 AcN/water/1% TFA, 230 nm). The reaction vessel was purged with nitrogen, Celite filter aid was added, stirred, filtered through a bed of Celite in a sintered glass funnel, the catalyst was washed with a 1:1:1 mixture of methanol/AcN/water, and the filtrate was concentrated to dryness. Toluene was added and then allowed to evaporate to dryness yielding 440 mg (81%) of a white solid - Compound E2-10(b). MS(FAB) = 1093.5 (parent ion)

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Alkylation of deprotected acyl hydrazide nucleus E2-10(b), where R is methyl, ethyl or n-propyl (E2-11)

E2-11

The following procedure illustrates a typical preparation for the alkylations.

Compound <u>E2-10(b)</u> (100 mg, 0.086 mmol) was dissolved or suspended in 10 ml

of dimethyl formamide. Two equivalents of the corresponding aldehyde (for formaldehyde use 5.2 mg (15 μl)) was added (for acetaldehyde use 7.6 mg (10 μl) and for propionaldehyde use 10 mg (12.5 μl), all are 0.173 mmol), followed by enough glacial acetic acid (approx. 3 - 4 drops) to make the mixture acidic. Sodium cyanoborohydride (11 mg, 0.173 mmol) was added and the mixture stirred at room temperature, overnight. The reaction was monitored by reverse-phase HPLC (C-18 Bondapak, AcN/water/1% TFA (55:35:10), 280 nm). The reaction was quenched with water to obtain a clear solution and a white gummy material in the bottom of the flask. The solvents were removed *in vacuo*, acetonitrile was added, filtered, and washed with ether to obtain a white powder (150 mg for methyl, 250 mg for ethyl, and 250 mg for propyl). All alkylated materials were isolated using preparative RP-HPLC(50:40:10 AcN/water/1% HCl, 230 nm for methyl and ethyl and 55:35:10 AcN/water/1% HCl, 230 nm for propyl).

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Yield: 64 mg for the methyl derivative  $\underline{E2-11(a)}$  having a MS(FAB) = 1107.54 (exact mass)

59 mg for the ethyl derivative  $\underline{\text{E2-11(b)}}$  having MS(FAB) = 1121.55 (exact mass)

73 mg for the propyl derivative  $\underline{E2-11(c)}$  having a MS(FAB) = 1135.57 (exact mass)

# Preparation of the product from coupling glycine to compound E2-7 (E2-12)

## E2-12

CBz-Glycine (2 g, 9.6 mmol) was dissolved in 25 ml of tetrahydrofuran,

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pentafluorophenol (2.2 g, 11.9 mmol)was added, followed by 1-[3-

(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride (EDAC) (2.2 g, 11.5

mmol). The mixture was stirred at room temperature under nitrogen. The

reaction was monitored by TLC (40% ethyl acetate/hexane, UV and CAM stain).

After approx. 1 hour, the solvent was removed in vacuo, the residue dissolved in

methylene chloride, the organic layer washed once with 1M sodium bisulfate

solution, thrice with 1N sodium hydroxide and finally once with brine. The

organic layer was dried over sodium sulfate, filtered and concentrated in vacuo to

yield 3.3 g (92%) of a light-pink solid. The material had consistent NMR data.

15 Compound E2-7 (700 mg, 0.45 mmol) was dissolved in 15 ml of

tetrahydrofuran. The active ester from above (504 mg, 1.3 mmol) was added, plus

a few drops of triethylamine. The mixture was heated near reflux for approx. 4

hours, then cooled to room temperature and allowed to stir overnight. The

volatiles were removed in vacuo, ether was added to the residue, washed twice

with 1N sodium hydroxide, once with brine, dried over sodium sulfate, filtered,

and concentrated *in vacuo* to yield 1.1 g of a white foam. Flash silica gel column cleanup (20 🗷 30% ethyl acetate/hexane) yielded 0.75 g (95%) of a white foam. The material had satisfactory NMR data. MS(FAB): 1727.9 (parent ion)

The silyl and BOC protecting groups were removed following the previous procedure for preparation of compound  $\underline{E2-9}$ . The reaction was monitored by RP-HPLC (25:65:10 AcN/water/1% TFA, 230 nm). Yield = 490 mg of a white powder

The terphenyl side chain was coupled to the above product according to the method for preparation of compound E2-10(b). Reacting amounts: 490 mg of the above compound, 260 mg (0.544 mmol) of the terphenyl active ester, 10 ml of dimethyl formamide, and enough diisopropylethylamine to make the reaction basic. The reaction was followed by RP-HPLC (25:65:10 AcN/water/1% TFA for starting material, 230 nm and 60:30:10 AcN/water/1% TFA for product, 280 nm). The solvent was removed under high vacuum to yield a white, gummy residue. Trituration from ether (2 washes) yielded 800 mg of a pale-yellow powder.

The above product was subjected to the same hydrogenolysis conditions as that for compound <u>E2-10(a)</u>. Reagent amounts: 400 mg of 10% palladium on charcoal, 100 ml of glacial acetic acid. Overnight reaction yielded 600 mg of an off-white solid. The product was isolated using RP-HPLC (AcN/water/1% TFA (50:40:10), 280 nm).

Yield = 188 mg of a white powder. NMR data was consistent with structure  $\underline{E2}$ - $\underline{12}$ . MS(FAB) = 1150.54 (exact mass)

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Preparation of the product from coupling diaminopropionic acid to compound E2-7 (E2-13)

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### E2-13

Following the above procedure for the preparation of compound E2-12, the active ester of L-di-CBz-diamino-propionic acid was prepared. Reacting stoichiometries were: L-di-CBz-diaminopropionic acid - 581 mg (1.56 mmol), pentafluorophenol - 344 mg (1.87 mmol), 1-[3-(dimethylamino) propyl]-3-ethylcarbodiimide hydrochloride (EDAC) - 360 mg (1.87 mmol), tetrahydrofuran - 12 ml. TLC system: chloroform/methanol/glacial acetic acid, 75:25:drop, for starting material; 40% ethyl acetate/hexane, TDM stain for product. Same workup yielded 720 mg (86%) of a white solid. The material had satisfactory NMR data.

Again, following the above procedure for coupling the active ester (710 mg, 1.32 mmol) to compound E2-7 (800 mg, 0.52 mmol) and refluxing for 2 days, standard workup yielded 1.1 g of a white foam. Flash silica gel column chromatography (100 g of silica, 20% ethyl acetate/hexane) yielded 0.56 g (57% yield) of a white foam. MS(FAB) = 1891 (parent ion)

The silyl and BOC protecting groups were removed following the previous procedure for preparation of compound <u>E2-9</u>. The reaction was monitored by RP-HPLC (50:40:10 AcN/water/1% TFA, 230 nm). Yield = 550 mg

The terphenyl side chain was coupled to the above product according to the method for preparation of compound E2-10(b). Reacting amounts: 550 mg of the above compound, 239 mg (0.5 mmol) of the terphenyl active ester, 10 ml of dimethylformamide, and enough diisopropylethylamine to make the reaction basic. The reaction was followed by RP-HPLC (70:20:10 AcN/water/1% TFA for starting material, 230 nm and 40:50:10 AcN/water/1% TFA for product, 280 nm). The solvent was removed under high vacuum to yield a white, gummy residue. Trituration from ether (2 washes) yielded 850 mg of an off-white powder.

The above product was subjected to the same hydrogenolysis conditions as that for compound  $\underline{E2-10(a)}$ . Reagent amounts: 400 mg of 10% palladium on charcoal, 100 ml of glacial acetic acid. Overnight reaction yielded 580 mg of an off-white solid. The product was isolated using RP-HPLC (gradient AcN/water/1% TFA (45/45/10  $\Rightarrow$  55/10 elution scheme, 280 nm). Two separate products were isolated, both having the same molecular weight. It was never determined what the relative stereochemistries of the two compounds were. The yield of one  $\underline{E2-13}$  isomer was 76 mg and the

yield of the other  $\underline{E2-13}$  isomer was 116 mg. MS(FAB) = 1179.6 (parent ion) for both

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Table 2 summarizes the activity data for compounds E2-9 through E2-13 in comparison with the comparative semi-synthetic Echinocandin compound  $\underline{C1}$ . The same testing procedures were used as described in Example 1 above.

Table 2						
		Minimal Inhibitory Concentration (MIC) μg/ml				
Example No.	C. albicans	C. parapsilosis	A. fumigatus	Histoplasma capsulatum		
Comparative C1	0.01	0.156	0.02	0.01		
E2-10(a)	0.625	>20	0.625	5.0		
E2-10(b)	0.01	0.156	0.156	0.156		
E2-11(a) R=methyl TFA salt	0.005	0.156	0.039	0.78		
E2-11(a) R=methyl HCl salt	0.001	0.156	0.078	0.02		
E2-11(b) R=ethyl HCl salt	0.001	0.156	0.312	0.02		
E2-11(c) R=n-propyl HCl salt	0.312	1.25	0.156	0.312		
E2-12	0.01	0.312	0.156	0.039		
E2-13 (isomer 1)	0.078	2.5	0.625	1.25		
E2-13 (isomer 2)	0.156	20	1.25	10		

Example 3 further illustrates the insertion of a new unit to yield an analog of an Echinocandin-type compound.

# Example 3

CbzNHCH<sub>2</sub>CH<sub>2</sub>SH was prepared as described in I. Shinkai, T. Liu, R. Reamer, M. Sletzinger, Synthesis, 924, 1980.

N-BOC-O-toluenesulfonyl serine methyl ester was prepared as described in N. A. Sasaki, C. Hashimoto, P. A. Potier, Tetrahedron Lett., 28, 6069-6072, 10 1987.

Preparation o(R)-2-[(tert-Butoxy-carbonyl)amino]-3-[(2'-N-benzyloxycarbonyl amino)ethanethio]methyl propiolate (E3-1)

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E3-1

Sodium hydride (72 mg, 1.8 mmol, 60% suspension in mineral oil) was triturated with hexanes under a nitrogen atmosphere in a 3-necked round bottomed flask. The flask was placed in a 0°C bath and a solution of CbzNHCH-2CH<sub>2</sub>SH (470 mg, 1.87 mmol, 85% pure) in DMF (5 ml) was added. The resulting mixture was stirred at 0°C for 20 min which resulted in a colorless solution. N-BOC-O-toluenesulfonyl serine methyl ester

(671 mg, 1.8 mmol) was added as a solid and washed into the flask with an additional 2 ml of DMF. The resulting mixture was stirred at 0°C for 3 hours then poured into water and extracted twice with ethyl acetate. The combined organic extracts were washed with water, 1N sodium hydroxide solution, water, and brine, then dried over MgSO<sub>4</sub> and concentrated *in vacuo* to give 900 mg of an oil. Radial chromatography eluting with 25% -> 50% ethyl acetate in hexanes gave 570 mg 76% of the desired (R)-2-[(tert-Butoxy-carbonyl)amino]-3-[(2'-N-benzyloxycarbonyl amino)ethanethio]methyl propiolate.

20 Anal. calculated for  $C_{19}H_{28}N_2O_6S$ , C: 55.32, H: 6.84, N: 6.79; Found C: 55.26, H: 6.95, N: 6.94.  $\lceil \alpha \rceil_D$  -1.9° (c=10).

<u>Preparation of (R)-2-[(tert-Butoxycarbonyl)amino]-3-[(2'-N-benzyloxycarbonyl amino)ethanethio]propiolic acid. (E3-2)</u>

E3-2

Compound E3-1 (520 mg, 1.26 mmol)in dioxane (3 ml) was treated with 0.5M LiOH solution (3 ml) and stirred at room temperature overnight. The dioxane was removed *in vacuo* and the residue partitioned between 1N hydrochloric acid solution and ethyl acetate. The organic extract was washed with brine, dried over MgSO<sub>4</sub> and concentrated *in vacuo* to give 500 mg of a colorless oil corresponding to compound E3-2. Anal. calculated for C<sub>18</sub>H<sub>26</sub>N<sub>2</sub>O<sub>6</sub>S+0.4H<sub>2</sub>O, C: 53.29, H: 6.65, N: 6.90; Found C: 53.61, H: 6.82, N: 6.85. [α]<sub>D</sub> -0.9° (c=10). MS: (m+1) 399.

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Preparation of (R)-2-[(tert-Butoxycarbonyl)amino]-3-[(2'-N-benzyloxycarbonyl amino)ethanesulfonyl]propiolic acid

(E3-3)

<u>E3-3</u>

Compound E3-2(0.95g, 2.38 mmol) was dissolved in MeOH (15 ml) and cooled to  $0^{\circ}$ C. A solution of Oxone® (1.77g, 5.7 mmol) in water (15 ml) was added and the resulting mixture stirred at  $0^{\circ}$ C for 1 hour then at room temperature overnight. The MeOH was removed *in vacuo* and the residue partitioned between ethyl acetate and water. The aqueous phase was extracted several more times with ethyl acetate. The combined organic extracts were washed with brine, dried over MgSO<sub>4</sub> and concentrated *in vacuo* to give 800 mg (78%) of a colorless foam corresponding to compound E3-3. Anal. calculated for  $C_{18}H_{26}N_2O_8S$ , C: 50.22, H: 6.09, N: 6.51. Found: , C: 50.13, H: 5.86, N: 6.45.  $[\alpha]_D$  -7° (c=10). MS: (m+1) 431.

# Preparation of Compound E3-4

E3-4

To a solution of Compound E3-3(160 mg, 0.37 mmol) and N-hydroxysuccinimide (43 mg, 0.37 mmol) in dry THF (5 ml) was added dicyclohexylcarbodiimide (76 mg, 0.37 mmol) and an additional 2 ml of THF. The mixture was stirred at room temperature for 3 hours then cooled to 0°C to help precipitate dicyclohexyl urea. In the meantime, the DiBOC CBZ silyl pentapeptide I-6(565 mg, 0.337 mmol) was dissolved in absolute ethanol (10 ml) and glacial acetic acid (95 ml), degassed and then 10% Pd/C (160 mg) was added to the mixture. The mixture was stirred under an atmosphere of H<sub>2</sub> (balloon pressure) for 3 hrs. The catalyst was removed by filtration and the filtrate concentrated *in vacuo* to a thick oil. THF (10 ml) and acetic acid (1 ml) were added and the solvents again removed *in vacuo* to remove all residual ethanol. The above prepared NHS active ester of compound

E3-3 was filtered directly into the flask containing the deblocked pentapeptide through a sintered glass funnel, washing the precipitated DCU with an additional 3 ml of THF. The resulting solution was made basic to Litmus paper by dropwise addition of triethyl amine. The mixture was stirred at room temperature for an additional 3 hours, then diluted with ethyl acetate, washed with saturated sodium bicarbonate solution, and brine, then dried over MgSO<sub>4</sub> and concentrated *in vacuo*. Flash chromatography eluting with 7:2:1 hexane:ethyl acetate:methylene chloride gave the desired coupling product E3-4 as a mixture of isomers. The less polar isomer yielded 140 mg having a MS = 1950.9(m+). Mixed fractions: 224 mg. The more polar isomer yielded 155 mg having a MS = 1951.9 (m+1). Total yield 519 mg, 78%.

This reaction was repeated on a 0.6 mmol scale to give 400 mg of the desired product as a mixture of isomers. Anal. calculated for C<sub>93</sub>H<sub>168</sub>N<sub>8</sub>O<sub>22</sub>SSi<sub>6</sub>, C: 57.25, H: 8.68, N: 5.75; Found C: 57.55, H: 8.63, N: 5.79.

15 (The stereocenter of acid  $\underline{E3-3}$  racemized in the coupling reaction.)

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# Preparation of Compound E3-5

E3-5

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A solution of compound E3-4 (390 mg, 0.2 mmol, mixture of diastereomers) in absolute ethanol (10 ml) and glacial acetic acid (10 ml) was degassed and then treated with 10% Pd/C (390 mg). The mixture was stirred under an atmosphere of H<sub>2</sub> (balloon pressure) for 2 hrs. The catalyst was removed by filtration through Celite and the filtrate concentrated *in vacuo* being careful to leave some acetic acid present. The residue was diluted with diethyl ether (175 ml) and triethylamine was added until the mixture was basic to Litmus paper (approximately 1 ml was required). The resulting solution was stirred at room temperature overnight, then washed with 0.1N hydrochloric acid solution, brine, dried over MgSO<sub>4</sub>, and concentrated *in vacuo* to give a foam. Flash chromatography on silica gel eluted with 30% ethyl acetate in hexanes gave the desired ring closed material E3-5. The less polar isomer yielded 105 mg of a

foam having a MS: (m+) 1599.9. The more polar isomer yielded 110 mg of a foam having a MS = 1599.8(m+). Total yield was 215 mg, 67%.

### Preparation of Compound E3-6

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E3-6

The more polar isomer of compound E3-5 (160 mg, 0.1 mmol) was dissolved in trifluoroacetic acid (5 ml) that had been cooled to 0°C. After 30 min water (0.5 ml) was added and the mixture stirred an additional 30 min at 0°C. The mixture was concentrated *in vacuo*, the residue dissolved in THF and concentrated *in vacuo*. The residue was dissolved in THF (3 ml), 1N hydrochloric acid solution (1 ml) was added, and the mixture placed in the refrigerator overnight. The solvent removed *in vacuo* then additional THF was added and the mixture reconcentrated to azeotrope off the water. This treatment provided a white solid. This white solid was dissolved in DMF(5 ml) and the terphenylhydroxybenzotriazole active ester (58 mg, 0.12 mmol) was added followed by triethylamine (60  $\square$ 1, 0.4 mmol). The resulting solution was stirred at

room temperature overnight, then the solvent was removed *in vacuo*. The residue was purified by preparative RP-HPLC (step gradient 50%-70% AcN in water over 45 min). The major peak was analyzed by HPLC (C-18 u-bondpak column, 60% AcN, 0.1% TFA in water) and the fractions containing the material which eluted at 4 min were combined and freeze dried to give 65 mg (56% yield) of  $\underline{E3-6}$  as a white powder having a MS = 1156.6(m+).

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E3-7

In analogous fashion the less polar isomer was converted to <u>E3-7</u> (material eluted in 7 min under the above described analytical conditions) to give 13 mg (11% yield) of the desired E3-7.

FAB MS (M), calculated for  $C_{58}H_{74}N_7O_{16}S$  1156.4913; found = 1156.4924.

The following set of examples illustrate the further cleavage of the linear pentapeptide to a tetrapeptide and subsequent insertion of new units onto the linear peptide chain prior to closing.

## Example 4

## E4-1

To a solution of Compound I-7 (732 mg, 1.1 mmol) in anhydrous THF (25 ml) was added PyBOP® (572 mg, 1.1 mmol), N-Cbz-L-valine (304 mg, 1.21 mmol), diisopropylethyl amine (0.57 ml, 3.3 mmol), and anhydrous DMF (1.0 ml) to dissolve remaining solids. The solution was stirred at room temperature for 2 hrs. followed by removal of the solvents *in vacuo*. The residue was dissolved in anhydrous THF (40 ml), followed by addition of t-Butyldimethylsilylchloride (1.66 g, 11.0 mmol) and imidazole (750 mg, 11.0 mmol). The solution was stirred at room temperature for 18 hrs. The reaction was concentrated *in vacuo* and the residue was dissolved in Et<sub>2</sub>O, washed twice with 0.1N HCl, dried over MgSO<sub>4</sub>, and the solvent removed *in vacuo*. Flash chromatography (eluting with 35% EtOAc/hexane) gave 690 mg of product, 46% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure E4-1. FAB MS (M<sup>+</sup>)= 1356.

**E4-2** 

To a solution of Compound <u>E4-1</u> (700 mg, 0.51 mmol) in anhydrous THF (0.6 ml) and acetonitrile (4 ml) was added N-t-Boc anhydride (0.24 ml, 1.03 mmol) and dimethylaminopyridine (7 mg, 0.05 mmol). The solution was stirred for 3 hrs. Following removal of the solvents *in vacuo*, flash chromatography (eluting with 17% EtOAc/hexane) gave 307 mg of product, 38% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure <u>E4-2</u>. FAB MS (M<sup>+</sup> of free base)= 1556.

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<u>E4-3</u>

To a solution of 5% Pd/C (150 mg) in EtOH (15 ml) and AcOH (15 ml) under an  $N_2$  atmosphere was added Compound E4-2 (307 mg, 0.20 mmol). The solution was purged/filled with  $H_2$  (x8) and subjected to constant  $H_2$  pressure for 2.0 hrs., then filtered over Celite to remove the catalyst. The solution was concentrated *in vacuo* to remove solvents, the residue dissolved in anhydrous THF (30 ml), followed by addition of  $\alpha$ -N-t-Boc- $\gamma$ -N-Cbz-L-ornithine (80 mg, 0.22 mmol), PyBOP® (103 mg, 0.20 mmol), and diisopropylethyl amine (0.10 ml, 0.59 mmol). The reaction was stirred at room temperature for 18 hrs., followed by removal of the solvent *in vacuo*. Flash chromatography (eluting with 30% EtOAc/hexane) gave 284 mg of a white solid, 81% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure E4-3. FAB MS (M<sup>+</sup> of free base)= 1771.

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## E4-4

To an N<sub>2</sub> purged solution of Compound E4-3 (279 mg, 0.16 mmol) in EtOH (13 ml) and AcOH (12 ml) was added 5% Pd/C (150 mg). The reaction was purged/filled with H<sub>2</sub> (x10) and left under constant H<sub>2</sub> pressure for 2 hrs., followed by removal of the catalyst by filtration over Celite and removal of solvents in vacuo. The resulting oil was dissolved in Et<sub>2</sub>O (75 ml) followed by addition of triethylamine (5 ml, 35.9 mmol). After 18 hrs., the reaction was concentrated in vacuo, and flash chromatography (eluting with 34% EtOAc/hexane) gave 96 mg of product, 43% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure E4-4. FAB MS (M<sup>+</sup> of free base)= 1420.

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E4-5

A solution of Compound <u>E4-4</u> (93 mg, 0.07 mmol) in ice cold TFA (2 ml) was placed in a 0°C freezer for 2 hrs., followed by addition of ice cold water (2 ml) and was then stirred in an ice bath for 2 hrs. The solution was concentrated *in vacuo* to yield 66 mg white solids, which were dissolved in THF (1.5 ml) and HCl (1.5 ml, 1.0N) and stirred at room temperature for 16 hrs. Toluene was added and the solution was concentrated *in vacuo* three times to assist in removal of TFA, giving 44.5 mg of an off-white solid, 91% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure <u>E4-5</u>. FAB MS (M<sup>+</sup> of free base)= 748.

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**E4-6** 

To a solution of Compound E4-5 (44 mg, 0.06 mmol) in anhydrous DMF (2 ml), was added diisopropylethyl amine (0.03 ml, 0.18 mmol) and the hydroxybenzotriazole active ester of the terphenyl side chain (34 mg, 0.07 mmol). The solution was stirred at room temperature for 40 hrs. Solvent was removed *in vacuo* and the residue was treated with Et<sub>2</sub>O (10 ml), sonicated, and a brown solid was isolated by filtration. RP-HPLC (eluting with 30-70% ACN/0.1% TFA/H<sub>2</sub>O) and freeze drying gave 18.7 mg of a white solid, 29% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure E4-6. FAB MS (M<sup>+</sup> of free base)= 1090.

# Example 5

Example 5 further exemplifies the insertion of new units onto a tetrapeptide chain followed by ring closure to produce a new cyclic peptide Echinocandin-type structure.

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<u>E5-1</u>

To a solution of tetrapeptide <u>I-7</u> (700 mg, 1.05 mmol) in anhydrous THF (20 ml) was added PyBOP® (547 mg, 1.05 mmol), N-Cbz-L-tyrosine (364 mg, 1.16 mmol), N,N-diisopropylethylamine (0.55 ml, 3.15 mmol). The solution was stirred at room temperature for 3 hrs. followed by removal of the solvents *in vacuo*. Used directly in the next reaction.

<u>E5-2</u>

The above residue <u>E5-1</u> was dissolved in anhydrous DMF (10 ml), followed by addition of t-Bu-dimethylsilyl chloride (1.90 g, 12.6 mmol) and

imidazole (860 mg, 12.6 mmol) and the solution was stirred at room temperature for 18 hrs. The reaction was concentrated *in vacuo* and the residue was dissolved in Et<sub>2</sub>O, washed twice with cold 0.1N HCl, once with H<sub>2</sub>O, 10% aq. sodium bicarbonate, brine and dried over MgSO<sub>4</sub>, filtered and the solvent removed *in vacuo* to give 2.0 g crude oil. Purified by radial chromatography (eluting with 35% EtOAc/hexanes) to give 700 mg (43% yield) of product. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure E5-2. FAB MS (M<sup>+</sup>)= 1534.

$$\begin{array}{c} \begin{array}{c} \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \end{array}$$

<u>E5-3</u>

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To a solution of Compound E5-2 (700 mg, 0.46 mmol) in anhydrous THF (10 ml) was added N-t-Boc anhydride (0.41 ml, 1.78 mmol) in portions and dimethylaminopyridine (7 mg, 0.05 mmol), and the solution was stirred for 3 hrs. at ambient temperature. Following removal of solvents *in vacuo*, purification by radial chromatography (eluting with 20% EtOAc/hexanes) gave 460 mg of product, 58% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure

E5-3. FAB MS ( $M^+$  of free base)= 1735.

<u>E5-4</u>

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To a solution of 10% Pd/C (270 mg) in EtOH (10 ml) and AcOH (10 ml) under an  $N_2$  atmosphere was added Compound E5-3 (460 mg, 0.27 mmol). The solution was purged/filled with  $H_2$  (x4) and subjected to constant  $H_2$  pressure for 2.0 hrs. at ambient temperature, then filtered over Celite to remove catalyst. The solution was concentrated *in vacuo* to remove solvents, the residue dissolved in anhydrous THF (30 ml), followed by addition of  $\alpha$ -N-t-Boc- $\gamma$ -N-Cbz-L-ornithine-N-hydroxysuccinimide ester (172 mg, 0.37 mmol), and triethylamine to pH 8( $\approx$ 5 ml). The reaction was stirred at room temperature for 3 hrs. Diluted the reaction with ether and washed with saturated sodium bicarbonate, 0.1N HCl, saturated sodium bicarbonate, and dried over magnesium sulfate. Filtered and concentrated *in vacuo*. Radial chromatography (eluting with 30% EtOAc/hexanes) gave 280 mg of a white solid, 54% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure E5-4. FAB MS (M<sup>+</sup> of free base)= 1949.

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<u>E5-5</u>

To an N<sub>2</sub> purged solution of Compound E5-4 (280 mg, 0.14 mmol) in EtOH (10 ml) and AcOH (5 ml) was added 10% Pd/C (200 mg). The reaction was purged/filled with H<sub>2</sub> (x4) and left under constant H<sub>2</sub> pressure for 2 hrs., followed by removal of the catalyst by filtration over Celite. The resulting oil was dissolved in Et<sub>2</sub>O (10 ml) followed by addition of triethylamine (4 ml, 28.7 mmol). After 18 hrs., the reaction was concentrated *in vacuo*, and purified by radial chromatography (eluting with 40% EtOAc/hexanes) gave 130 mg of product, 57% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure E5-5. FAB MS (M<sup>+</sup> of free base)= 1597.

<u>E5-6</u>

A solution of Compound  $\underline{E5-5}$  (128 mg, 0.08 mmol) in ice cold TFA (2 ml) was placed in 0°C freezer for 2 hrs., followed by addition of ice cold water (2 ml) and was then stirred in an ice bath for 2 hrs. The solution was concentrated *in vacuo*, then dissolved in THF (1.5 ml) and 1N HCl (1.5 ml) and stirred at room temperature for 16 hrs. Toluene was added and the solution was concentrated *in vacuo* three times, giving 70 mg of a dihydrochloride white solid, 100% yield. The  $^{1}$ H NMR (300 MHz) spectrum was consistent with the structure  $\underline{E5-6}$ . FAB MS calculated for M+H  $C_{39}H_{54}N_{7}O_{12}$ = 812.3830; found = 812.3837.

E5-7

To a solution of Compound  $\underline{E5-6}$  (65 mg, 0.08 mmol) in anhydrous DMF (5 ml), was added N,N-diisopropylethylamine (2.0 ml, 11.5 mmol) and the hydroxybenzotriazole active ester of the terphenyl side chain (50 mg, 0.11 mmol), and the solution was stirred at room temperature for 18 hrs. Solvent was removed in vacuo and the residue was treated with  $Et_2O$  (10 ml), sonicated, and a beige solid was isolated. The methanol soluble portion was purified by RP HPLC (Waters Bondapak C-18, eluting with 58% AcN/0.1% TFA/H<sub>2</sub>O at a flow of 20 ml/min) and freeze drying gave 35 mg of a white solid, 38% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure  $\underline{E5-7}$ . FAB MS calculated for M+H C<sub>63</sub>H<sub>76</sub>N<sub>7</sub>O<sub>14</sub> = 1154.5450; found = 1154.5458.

Example 6 illustrates the introduction of a water solubilizing group onto the tetrapeptide intermediate prior to cyclization.

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#### Example 6

#### E6-1

N-α-Boc-L-α,β- diaminopropionic acid (0.5 g, 2.45 mmol)(available from Bachem) and 1,3-Bis(benzyloxycarbonyl)-2-methyl-2-thiopseudourea (0.88 g, 2.45 mmol) were combined in 15 ml of anhydrous DMF. Triethylamine (1.0 ml, 7.3 mmol) was added and stirred for 3 days at ambient temperature. The reaction was diluted with 100 ml of 0.1 N NaOH and extracted into ether. The aqueous layer was then acidified with cold saturated citric acid and extracted with ethyl acetate (3 X 200 ml). The combined organics were dried over MgSO<sub>4</sub>, filtered and concentrated to a quantitative yield of a thick colorless oil. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with structure <u>E6-1</u>. FD MS (M<sup>+</sup> of free base)= 515.

E6-2

To the above acid <u>E6-1</u> (310 mg, 0.60 mmol) in anhydrous THF (10 ml) was added N-hydroxysuccinimide (69 mg, 0.60 mmol) and DCC (123 mg, 0.60

mmol). A white precipitate began to form after about 1 hr. The reaction was allowed to stir overnight at ambient temperature. The reaction was filtered and the crude solution used directly in the next coupling.

E6-3

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To a solution of the above activated ester <u>E6-2</u> (366 mg, 0.60 mmol) in anhydrous THF (10 ml) and ether (10 ml) was added Compound <u>E5-3</u>(920 mg, 0.60 mmol) and triethylamine (3 ml). The solution was allowed to stir overnight (18 hrs.) at ambient temperature. The reaction was diluted with ether and washed with saturated sodium bicarbonate (1 X 250 ml), 0.1 N HCl (1 X 250 ml), saturated sodium bicarbonate (1 X 250 ml), dried over magnesium sulfate, filtered and concentrated to 1.0 g crude white solid. The product was purified by radial chromatography (eluting with 30/70 ethyl acetate/hexanes) to give 550 mg (46% yield) of a white solid. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with structure <u>E6-3</u>. FAB MS (M<sup>+</sup> of free base)= 2036.

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E6-4

To a solution of 5% Pd/C (150 mg) in EtOH (15 ml) and AcOH (15 ml) under an  $N_2$  atmosphere was added Compound  $\underline{E6-3}$  (550 mg, 0.27 mmol). The solution was purged/filled with  $H_2$  (x4) and subjected to constant  $H_2$  pressure for 2.0 hrs., then filtered over Celite to remove the catalyst. The solution was concentrated *in vacuo* to remove the solvents, the residue dissolved in acetonitrile (10 ml) and ether (3 ml), followed by addition of 2 ml of triethylamine. The mixture was sonicated for 4 hours and the temperature allowed to reach  $40^{\circ}$ C. The reaction mixture was concentrated and purified by radial chromatography (eluting with 40/60 ethyl acetate/hexanes) to give 57 mg of the desired compound (14% yield). The  $^{1}$ H NMR (300 MHz) spectrum was consistent with the structure  $\underline{E6-4}$ . FAB MS ( $M^{+}$  of free base)= 1549.

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<u>E6-5</u>

To the above compound <u>E6-4</u> (77 mg, 0.05 mmol) was added 3 ml of neat trifluoroacetic acid while cooling to 0°C. After 45 minutes, 0.5 ml of water was added while maintaining the reaction at 0°C. After 30 minutes, the mixture was concentrated to a colorless oil. To this oil was added 2 ml of THF and 2 ml of 1N HCl and refrigerated overnight. Toluene was stripped from the mixture to give a quantitative yield of the free amine as the trihydrochloride salt. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure <u>E6-5</u>. FAB MS (M<sup>+</sup> of free base)= 764.

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E6-6

To a solution of Compound <u>E6-5</u> (50 mg, 0.06 mmol) in anhydrous DMF (5 ml), was added N,N-diisopropylethylamine (2 ml) and the hydroxybenzotriazole active ester of the terphenyl side chain (44 mg, 0.09 mmol), and the solution was stirred at room temperature for 18 hrs. The solvent was removed *in vacuo* and the residue was triturated with a mixture of acetonitrile and ether. The solid was dried to 25 mg of a crude white solid. The product was purified by RP HPLC on a Waters Bondapak C-18 column eluting with 55% AcN/0.1% TFA/H<sub>2</sub>O at a flow of 20 ml/min. The appropriate fractions were freeze dried to give 10.0 mg of a white solid, 18% yield. The <sup>1</sup>H NMR (300 MHz) spectrum was consistent with the structure <u>E6-6</u>. FAB MS calculated for (M+H) C<sub>57</sub>H<sub>72</sub>N<sub>9</sub>O<sub>14</sub> = 1106.5199; found = 1106.5185

Table 3 summarizes the activity data for compounds

E3-6, E4-6, E5-7 and E6-6 in comparison with the comparative semi-synthetic Echinocandin compound C1. The same testing procedures were used as described in Example 1 above.

Table 3

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	Minimal Inhibitory Concentration (MIC) μg/ml							
Example No.	C. albicans	C. parapsilosis	A. fumigatus	Histoplasma capsulatum				
Comparative C1	0.01	0.156	0.02	0.01				
E3-6	10	>20	20	10				
E3-7	>20	>20	>20	>20				
E4-6	1.25	>20	>20	>20				
E5-7	>20	>20	>20	>20				
E6-6	0.078	>20	>20	>20				

#### Example 7

Example 7 illustrates the formation of a cyclic heptapeptide from the intermediate linear pentapeptide

(I-6).

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Preparation of N-α-BOC-D-2,3-diaminopropionic acid (E7-1)

E7-1

To a 1:1 dimethylformamide:water solution (170 ml) of [bis(trifluoroacetoxy)iodo]benzene (12.89 g, 32.29 mmol, 1.5 equiv) was added N-α-BOC-D-asparagine (5 g, 21.53 mmol, 1 equiv). This solution stirred at room temperature for .5 h before pyridine (3.4 g, 43.06 mmol, 2 equiv) was added. After 18 h the reaction was concentrated *in vacuo* and the residue was redissolved in water before being washed with diethyl ether (2X, 50 ml). The aqueous layer

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was concentrated *in vacuo* and the crude product was recrystallized from hot acetonitrile to give <u>E7-1</u> (1.10 g, 25% yield).

<u>Preparation of N- $\alpha$ -BOC-D-2,3-diaminopropionic acid-N-CBZ-glycine dipeptide</u> (E7-2)

E7-2

An aqueous solution (12 ml) of N-α-BOC-D-2,3-diaminopropionic acid E7-1 (1.114 g, 5.45 mmol, 1 equiv) and NaHCO<sub>3</sub> (0.458 g, 5.45 mmol, 1 equiv) was stirred rapidly for 15 minutes until complete solvation. To this was added a 1,2-dimethoxyethane solution (22 ml) of N-CBZ-O-N-hydroxysuccinimide glycine ester. After stirring at room temperature for 18 hours the reaction was concentrated *in vacuo*. The residue was redissolved in water, acidified to pH 3 with 1N aqueous HCl, and partitioned between ethyl acetate and water. The aqueous layer was washed 3X with additional water before organics were combined, dried over MgSO<sub>4</sub>, and concentrated. The crude white foam was purified on reverse phase, C-18 column, preparative HPLC (gradient 5:95 AcN/0.01%TFA to 100%AcN elution scheme) to afford 1.46 g (3.69 mmol, 68 % yield) of E7-2.

# Preparation of N-α-BOC-D-2,3-diaminopropionic acid-N-CBZ-glycine dipeptide-O-NHS active ester (E7-3)

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E7-3

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To a 1,2-dimethoxyethane solution (40 ml) of E7-2 (1.40 g, 3.54 mmol, 1 equiv) and N-hydroxysuccinimide (0.448 g, 3.89 mmol, 1.1 equiv) cooled to 0°C was added dicyclohexylcarbodiimide (0.804 g, 3.89 mmol, 1.1 equiv). After stirring for 1 h at 0°C it was set in the refrigerator for 18 hours. The solution was then filtered and the filtrate was stripped to dryness and placed on high vacuum for 2 hours to give approximately 2 g of product (contained some DCU byproduct) which was used without further purification.

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# 5 <u>DiBOC silyl N(α)BOC-D-2,3-diaminopropionic acid-glycine-CBZ heptapeptide</u> (E7-4)

<u>E7-4</u>

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A solution of linear peptide intermediate  $\underline{\text{I-6}}$  (1.0 g, 0.598 mmol) in ethanol (5 ml) was added to a slurry of 10% Pd/C (250 mg) in 5 ml of ethanol followed by 10 ml of glacial acetic acid. The mixture was put under a balloon of  $H_2$  and after 1 hr the starting material was gone (TLC 25% ethyl acetate/hexane). The catalyst was removed by filtration through a Celite plug and the solution was carefully reduced (but not to dryness) under high vacuum keeping the temperature under  $40^{\circ}$ C. The resulting oil was dissolved in 25 ml of ether and a 10 ml

tetrahydrofuran solution of dipeptide active ester  $\underline{E7-3}$  (O-Suc-N $\alpha$ -BOC-D-2,3-diaminopropionic acid-N-CBZ-glycine) was added followed by excess triethylamine until the solution was basic to pH paper. After stirring for 16 hrs., the solution was extracted with saturated NaHCO3 solution followed by dilute HCl solution and then another portion of saturated NaHCO3 solution. The organic layer was dried over MgSO4 and reduced *in vacuo* to give 1.194 g of the crude product. Purification by flash chromatography (30% ethyl acetate/hexane) gave 0.674 g (59% yield) of coupled product  $\underline{E7-4}$ . FAB MS = 1916.5 (M+1)

#### 10 <u>Cyclization of E7-4 to BOC silyl cycloheptapeptide (E7-5)</u>

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E7-5

An ethanol/acetic acid solution (10 ml of each) of E7-4 (0.665 g, 0.34 mmol) with 10% Pd/C (200 mg) was placed under a balloon of hydrogen. After 1.5 hrs., TLC (30% ethyl acetate/hexane) indicated a complete reaction. The catalyst was removed by filtration through a plug of Celite and the solvent reduced *in vacuo* (but not to dryness) at 40°C until the residue was a thick oil.

This material was dissolved in ethyl ether (150 ml) and excess triethylamine ( $\sim$ 8 ml) was added. After 18 hrs., TLC (30% ethyl acetate/hexane) indicated one major product spot. The solvent was removed *in vacuo* and the residue was redissolved in ethyl acetate and washed several times with water. The organics were combined and dried over MgSO<sub>4</sub> and the solvent removed *in vacuo* to give 0.800 g of crude product. This was purified over a flash column (silica gel eluted with 30% ethylacetate/hexanes) to provide 293 mg (54% yield) of <u>E7-5</u> as a white solid. FAB MS = 1564.9

## 10 Removal of protecting groups and coupling of the side chain to give E7-6

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E7-6(a)

Compound E7-5 (288 mg, 0.181 mmol) was dissolved in trifluoroacetic acid (3 ml) and cooled to 0°C. After 0.5 hrs., water was added (0.5 ml) and the mixture was stirred for 0.6 hrs longer. The solvent was removed *in vacuo* and the residue was dissolved in 1N HCl (2 ml) and tetrahydrofuran (3 ml). This solution was stirred at room temperature for 1.5 hr after which time it was set in the refrigerator overnight. The solvent was removed under high vacuum giving a

foam residue which was dissolved in anhydrous dimethylformamide (3 ml). Terphenyl hydroxybenzotriazole active ester (108 mg, 0.276 mmol) and triethylamine (0.11 ml, 0.78 mmol) were added to the solution. After stirring overnight at room temperature the solvents were removed under high vacuum and the crude material (380 mg) was purified by preparative RP-HPLC (C-18 column eluted with a 50% AcN/0.01%TFA aqueous solution). Lyophilization of the pure fractions gave 97 mg (48% yield) of E7-6(a) as a white solid. FAB MS = 1121.5 (M) calc. for  $C_{58}H_{72}N_8O_{15} = 1121.21$ 

### 10 Preparation of E7-6(b)

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In a similar manner, N-α-BOC-L-asparagine was converted to E7-6(b).

E7-6(b)

The H-NMR data was consistent with the structure  $\underline{E7-6(b)}$ . MS(FAB) = 1121 (M+)

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#### Preparation of N-α-CBZ-D-2,3-diaminopropionic acid (E7-7)

E7-7

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Compound <u>E7-7</u> was prepared in a similar manner to N- $\alpha$ -BOC-diamino propionic acid <u>E7-1</u>. MS FAB (M+1) = 239

#### Preparation of N-α-CBZ-N-β-BOC-D-2,3-diaminopropionic acid (E7-8)

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E7-8

To a stirring solution of sodium hydroxide (148 mg, 3.69 mmol, 1.1 equiv) in water (5 ml) was added N-α-CBZ-diamino propionic acid E7-7. The reaction was stirred for 10 minutes before tert-butyl alcohol (4 ml) was added. The reaction was cooled to 0°C and di-tert-butyl dicarbonate (807 mg, 3.69 mmol, 1.1 equiv) was added slowly over 0.5 h. After stirring overnight at room temperature, the reaction was diluted with water (5 ml) and washed 3X with 10 ml ethyl ether. The organics were then combined and washed several times with saturated aqueous sodium bicarbonate. The aqueous layers were combined, cooled to 0°C, and acidified to pH 3 with aqueous potassium hydrogen sulfate (30 g in 200 ml stock solution). This cloudy solution was then extracted several times with ethyl

acetate. The organics were combined, dried over MgSO<sub>4</sub>, and concentrated *in* vacuo to give after overnight high vacuum 0.990 g (2.9 mmol, 87% yield) of E7-8. The  $^{1}$ H NMR was consistent with structure E7-8. MS FAB (M+1) = 339

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## Preparation of N-β-BOC-D-2,3-diaminopropionic acid (E7-9)

0 NH NH OH E7-9

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To an ethyl alcohol solution (20 ml) of N- $\alpha$ -CBZ-N- $\beta$ -BOC-D-2,3-diaminopropionic acid <u>E7-8</u> was added 10% palladium on carbon catalyst (approx. 200 mg). The mixture was placed under an H<sub>2</sub> atmosphere and stirred vigorously. Due to gel like formation, the reaction required additional ethyl alcohol (total volume of 75 ml) to facilitate easy stirring. After several hours, the reaction was filtered through a plug of Celite and then concentrated *in vacuo* to give 259 mg (1.27 mmol, 32 % yield) of <u>E7-9</u>.

<u>Preparation of N-β-BOC-D-2,3-diaminopropionic acid-N-CBZ-glycine dipeptide</u>
(E7-10)

E7-10

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Compound  $\underline{E7-10}$  was prepared in a similar manner as N- $\alpha$ -BOC-D-2,3-diaminopropionic acid-N-CBZ-glycine dipeptide  $\underline{E7-2}$ . The <sup>1</sup>H NMR was consistent with the structure  $\underline{E7-10}$ .

10 <u>Preparation of N-β-BOC-D-2,3-diaminopropionic acid-N-CBZ-glycine dipeptide-</u> O-NHS active ester (E7-11)

E7-11

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Compound  $\underline{E7-11}$  was prepared in a similar manner as N- $\alpha$ -BOC-D-2,3-diaminopropionic acid-N-CBZ-glycine dipeptide-O-NHS active ester  $\underline{E7-3}$ .

# <u>Preparation of DiBOC silyl N(β)BOC-D-2,3-diaminopropionic acid-glycine-CB2</u> heptapeptide (E7-12)

E7-12

Compound  $\underline{E7-12}$  was prepared in a similar manner as DiBOC silyl  $N(\alpha)BOC-D-2,3$ -diaminopropionic acid-glycine-CBZ heptapeptide  $\underline{E7-4}$ . MS

FAB(M+1) = 1917

# Preparation of BOC silyl cycloheptapeptide (E7-13)

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E7-13

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Compound  $\underline{E7-13}$  was prepared in a similar manner as BOC silyl cycloheptapeptide  $\underline{E7-5}$ .

#### Preparation of Cycloheptapeptide E7-14(a)

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E7-14(a)

Compound E7-14 was prepared in a similar manner as cycloheptapeptide

E7-6. MS FAB (M) = 1121.6

#### Preparation of Cycloheptapeptide E7-14(b)

In a similar manner as <u>E7-14(a)</u>, <u>E7-14(b)</u> was prepared from N- $\alpha$ -CBZ-L-2,3-diamino propionic acid.

#### E7-14(b)

The H<sup>1</sup>-NMR data was consistent with structure E7-14(b). MS(FAB) = 1121 (M+)

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#### Example 8

<u>Preparation of (-L-)-( $\alpha$ )-N-CBZ-( $\beta$ )-N-trifluoroacetyl 2,3-diaminopropionic acid</u> (E8-1)

The procedure of Curphey et al., *J. Org. Chem.*, 44, 2805, (1979) was utilized as follows. A suspension of (-L)-( $\alpha$ )-N-CBZ-2,3-diaminopropionic acid (2.0 g, 8.39 mmol) and triethylamine (0.84 g, 8.39 mmol) in methanol (10 ml) at ambient temperature was treated with ethyl trifluoroacetate (1.49 g, 10.49 mmol) and the mixture stirred for 48 hrs. The resulting solution was diluted with methanol (5 ml), cooled to 0°C, and treated with Dowex 50W resin(3.30 g). After stirring for 10 min., the suspension was filtered and the filtrate concentrated *in vacuo* to produce 2.74 g of a white solid (98% yield) that was used without further purification. H<sup>1</sup> NMR data was consistent with the structure E8-1. MS(FD) = 334 (M+)

#### Preparation of the N-hydroxysuccinimide ester (E8-2) from E8-1

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#### E8-2

To a solution of E8-1 (1.20 g, 3.59 mmol) and N-hydroxysuccinimide (0.45 g, 3.95 mmol) in 1,2-dimethoxyethane (20 ml) at 0°C was added N,N'-dicyclohexylcarbodiimide (0.81 g, 3.95 mmol). The mixture was stirred at cold bath temperature for 2 hr followed by overnight storage in the refrigerator. Filtration of the suspension and subsequent concentration of the filtrate gave a crude solid product which was recrystallized from ethyl acetate/hexanes to produce 0.78 g of a crystalline solid (50% yield, one crop). The H $^1$  NMR data was consistent with structure E8-2. MS(FD) = 431 (M+)

Preparation of dipeptide (E8-3) from amino acid active ester E8-2

E8-3

(-L-)-(α)-N-BOC-2,3-diaminopropionic acid (0.52 g, 2.55 mmol) was
dissolved in aqueous sodium bicarbonate solution (prepared from dissolving 0.22 g, 2.55 mmol of sodium bicarbonate in 10 ml of water). This solution was added to a solution of active ester E8-2 (1.1 g, 2.55 mmol) in 1,2-dimethoxyethane (23 ml) and the mixture stirred for 24 hrs. After concentration *in vacuo* to remove 1,2-dimethoxyethane, the residual suspension was adjusted to pH 5 with 1N aqueous citric acid, then extracted with ethyl acetate (2x). The combined organic extracts were washed successively with water and brine, dried over MgSO<sub>4</sub> and reduced *in vacuo* to give 1.4 g of a crude foam. Trituration with methylene chloride gave 1.05 g of a flocculent solid (75% yield). Additional product in mother liquor was not recovered.

15 H<sup>1</sup> NMR data was consistent with structure E8-3. MS(negative ion electrospray) = 519 (M-H)

# Preparation of dipeptide active ester (E8-4) from dipeptide E8-3

#### E8-4

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To a solution of E8-3 (0.65 g, 1.24 mmol) and N-hydroxysuccinimide (0.16 g, 1.37 mmol) in tetrahydrofuran (5 ml) at 0°C was added N,N'dicyclohexylcarbodiimide (0.28 g, 1.37 mmol). The mixture was stirred at cold bath temperature for 2 hrs followed by overnight storage in the refrigerator. Filtration of the suspension and subsequent concentration of the filtrate gave 0.70 g of a crude foam (89% yield). H<sup>1</sup> NMR data was consistent with structure 10 E8-4. MS(FD) = 631 (M+)

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Preparation of DiBOC silyl (-L-)- $(\alpha)$ -N-BOC-2,3-diaminopropionic acid-(-L-)- $(\alpha)$ -N-CBZ- $(\beta)$ -N-trifluoroacetyl 2,3-diaminopropionic acid linear heptapeptide (E8-5)

E8-5

A solution of linear pentapeptide intermediate I-6

(2.0 g, 1.19 mmol) in ethyl acetate(10 ml) was added to a slurry of 10% Pd/C (400 mg) in ethyl acetate (15 ml) followed by 20 ml of glacial acetic acid. The mixture was put under a balloon of  $H_2$  and after 1 hr the starting material was gone. The catalyst was removed by filtration and the solution was carefully reduced under high vacuum keeping the temperature under 40°C. The resulting oil was dissolved in THF (15 ml) and the dipeptide active ester, N-(-L-)-( $\alpha$ )-CBZ-N-( $\beta$ )-trifluoroacetyl 2,3-diaminopropionic acid- N-(L)-( $\alpha$ )-BOC 2,3-diaminopropionic acid-Osu E8-4, was added followed by excess triethylamine until the solution was

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basic to pH paper. After stirring for 18 hrs., the solution was reduced *in vacuo* and the residue partitioned between ethyl ether and water. The ether layer was washed with saturated NaHCO<sub>3</sub> solution, followed by successive washings with water, 1N aqueous citric acid, water, saturated NaHCO<sub>3</sub> and brine. The organic layer was dried over MgSO<sub>4</sub> and reduced *in vacuo* to give 2.36 g of the crude product.

Purification by silica flash chromatography (25% ethyl acetate/hexane) gave 1.22 g of coupled product E8-5 as a foam (56% yield). H<sup>1</sup> NMR data was consistent with structure E8-5. MS(FAB) = 2041.5 (M+)

#### Cyclization of E8-5 to BOC silyl cycloheptapeptide (E8-6)

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E8-6

An ethyl acetate/acetic acid solution (20 ml each) of E8-5 (1.20 g, 0.58 mmol) with 10% Pd/C (290 mg) was placed under a balloon of hydrogen. After 1.5 hrs., TLC indicated deprotection was complete. The catalyst was removed by filtration and the filtrate concentrated *in vacuo* to a thick slurry. This material was

dissolved in ethyl ether(120 ml) and excess triethylamine was added until the solution was basic to pH paper(~5 ml). After 36 hrs., TLC indicated one major product. The solution was washed successively with water, 1N aqueous citric acid, water, and brine. The organic layer was dried over MgSO<sub>4</sub> and reduced *in vacuo* to give 1.14 g of crude product. Purification by silica flash chromatography (25% ethyl acetate/hexane) gave 0.69 g of E8-6 as a foam (70% yield). H<sup>1</sup> NMR data was consistent with structure E8-6. MS(FAB) = 1690.0(M+H)

10 Removal of protecting groups and coupling of the side chain to generate E8-7

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E8-7

A solution of E8-6 (0.69 g, 0.40 mmol) in trifluoroacetic acid (23 ml) at 0°C was stirred for 0.5 hr after which time water (2 ml) was added and the stirring continued for an additional 0.75 hr at 0°C. The solvent was removed *in vacuo* and the residue was dissolved in tetrahydrofuran (9 ml) and treated with 1N HCl (4 ml). This solution was stirred at ambient temperature for 1.25 hr and then refrigerated for 18 hr. HPLC showed one major product peak. Concentration *in* 

vacuo produced a residual foam which after dissolution in dimethylformamide (12 ml) was treated with the terphenyl hydroxybenzotriazole active ester (0.25 g, 0.52 mmol) and triethylamine (0.28 ml, 2.0 mmol). After stirring at ambient temperature for 17 hrs., the solvent was removed under high vacuum and the crude residue purified by preparative RP-HPLC (linear gradient 60%-100% AcN/0.1%TFA elution scheme) to produce 0.37 g of a white solid (75% yield). H<sup>1</sup> NMR data was consistent with structure E8-7. MS(FAB) = 1246.7(M+)

## Final deprotection of E8-7 to generate E8-8

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#### E8-8

To a solution of E8-7 (250 mg, 0.20 mmol) in methanol (12 ml) was added a solution of potassium carbonate (138 mg, 1.0 mmol) in water (6 ml), and the resulting mixture stirred at ambient temperature for 20 hrs. Solvent removal *in vacuo* followed by purification via preparative RP-HPLC (linear gradient 60-100% AcN/0.1%TFA elution scheme) gave 218 mg of a white solid (94% yield). H<sup>1</sup> NMR data was consistent with structure E8-8. MS(FAB) = 1150.6(M+)

#### Reductive alkylation of E8-8 to generate E8-9

E8-9

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To a solution of E8-8 (40 mg, 0.0347 mmol) in methanol (2 ml) at ambient temperature was added 1-methyl-4-piperidone (7.85 mg, 0.0694 mmol) and glacial acetic acid (2  $\mu$ l, 0.0347 mmol). The solution was treated with sodium cyanoborohydride (3.27 mg, 0.0520 mmol) and the mixture stirred for 16 hrs. After concentration *in vacuo*, the crude product was purified via preparative RP-HPLC (step gradient 40-100% AcN/0.1%TFA elution scheme) to yield 19 mg of a white solid (45%

yield). H<sup>1</sup> NMR data was consistent with structure  $\underline{E8-9}$ . MS(FAB) = 1247.6

(M+)

Table 4 summarizes the activity data for compounds

E7-6(a), E7-6(b), E7-14(a), E7-14(b), E8-8 and E8-9 in comparison with the comparative semi-synthetic Echinocandin compound C1 and Amphotericin B.

The same testing procedures were used as described in Example 1 above.

Table 4

1 able 4								
	Minimal Inhibitory Concentration (MIC) μg/ml							
Example No.								
	Candida albicans	Candida parapsilosis	Aspergillus fumigatus	Cryptococcus neoformans	Histoplasma capsulatum			
Comparative C1	0.01	0.156	0.02	>20	0.01			
Ampho B	0.078	0.038	0.312	0.039	0.039			
E7-6(a) (22 membered ring)	>20	>20	>20	>20	>20			
E7-6(b) (22 membered ring)	0.312	>20	5.0	>20	1.25			
E7-14(a) (21 membered ring)	0.312	>20	>20	>20	0.312			
E7-14(b) (21 membered ring)	0.156	>20	>20	>20	0.078			
E8-8 (22 membered ring)	0.078	0.625	5.0	>20	0.312			
E8-9 (22 membered ring)	0.156	5.0	5.0	>20	0.625			

#### WE CLAIM:

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1. A process for modifying a cyclic peptide ring nucleus comprising the steps of:

- (i) providing a cyclic peptide compound comprising a peptide unit having a γ-hydroxyl group;
- (ii) opening the ring of said cyclic peptide compound to provide a first linear peptide wherein said peptide unit having a γ-hydroxyl group is the N-terminus peptide unit of said first linear peptide;
- (iii) cleaving-off said peptide unit having a γ-hydroxyl group to provide a second linear peptide;
- (iv) attaching at least one amino acid, dipeptide unit or synthetic unit to said second linear peptide to produce a third linear peptide;
- (v) cyclizing said third linear peptide to produce a modified cyclic peptide compound having a modified ring nucleus.

2. The process of Claim 1 wherein said amino acid, said dipeptide unit or said synthetic unit of step (iv) comprises a protected amino group.

- 3. The process of Claim 2 further comprising
- 20 (vi) deprotecting said protected amino group to provide a deprotected amino group;
  - (vii) acylating said deprotected amino group.
- 4. The process of Claim 1 or Claim 2 further comprising cleaving another peptide unit from said second linear peptide in step (iii) before attaching said at least one amino acid, dipeptide unit or synthetic unit in step (iv).

5. The process of Claim 1 wherein step (iii) is performed by adding trifluoroacetic acid or hydrochloric acid to said first linear peptide in an organic solvent.

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- 6. The process of Claim 5 wherein said organic solvent is selected from the group consisting of methylene chloride, toluene and dioxane.
- 7. The process of Claim 1 or 2 wherein a second amino acid, dipeptide or synthetic unit is attached to said third linear peptide in step (iv) prior to cyclizing in step (v).
  - 8. The process of Claim 4 wherein a second amino acid, dipeptide or synthetic unit is attached to said third linear peptide in step (iv) prior to cyclizing in step (v).
  - 9. The process of Claim 1 wherein said cyclic peptide compound is a cyclic hexapeptide.
- 20 10. The process of Claim 1 wherein said cyclic peptide compound is represented by the following structure:

wherein R is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, or heteroaryl group;  $R^1$  is -H or -OH;  $R^2$  is -H or -CH<sub>3</sub>;  $R^3$  is -H, -CH<sub>3</sub>, -CH<sub>2</sub>CONH<sub>2</sub> or -CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>;

- 5  $R^4$  is -H or -OH;  $R^5$  is -OH, -OPO<sub>3</sub>H<sub>2</sub>, or -OSO<sub>3</sub>H; and  $R^6$  is -H or -OSO<sub>3</sub>H.
  - 11. The process of Claim 1 wherein said modified cyclic peptide compound is a 19-, 20-, 21-, or 22-membered ring compound.
- 10 12. A compound prepared by the process according to Claim 10 and represented by formula I or II:

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wherein

R is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, or heteroaryl group;

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R<sup>2</sup> is -H or -CH<sub>3</sub>;

10 R<sup>3</sup> is -H, -CH<sub>3</sub>, -CH<sub>2</sub>CONH<sub>2</sub> or -CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>;

R<sup>4</sup> is -H or -OH;

 $R^5$  is -OH, -OPO<sub>3</sub>H<sub>2</sub>, or -OSO<sub>3</sub>H;  $R^6$  is -H or -OSO<sub>3</sub>H;  $R^7$  is -CH<sub>3</sub> or -H;

(Y) is represented by the following formula

NH-W B O | II | II | CH-C-

wherein

A is -(CH<sub>2</sub>)<sub>a</sub>- where a = 1-4, -CHR'-CHR''-(CH<sub>2</sub>)<sub>b</sub>- where R' and R'' are independently -H, -OH,  $C_6H_5O$ -, -SH,

-NH<sub>2</sub>,  $C_nH_{n+2}NH$ -,  $C_nH_{n+2}O$ -,  $C_nH_{n+2}S$ - or - $C_nH_{n+2}$  where n = 1-4 and b = 0-1, -(CH<sub>2</sub>)<sub>c</sub>-C(O)NH(CH<sub>2</sub>)<sub>d</sub>- where c = 1-2 and d = 1-2, -N=CH-(CH<sub>2</sub>)<sub>e</sub>- where e = 0-2, -NR'''(CH<sub>2</sub>)<sub>f</sub>- where R''' is -H, -C(O)CH<sub>2</sub>NH<sub>2</sub>,

-C(O)CH(NH<sub>2</sub>)CH<sub>2</sub>NH<sub>2</sub> or - $C_nH_{n+2}$  where n = 1-4 and f = 1-3, -(CH<sub>2</sub>)<sub>g</sub>-SO<sub>2</sub>-(CH<sub>2</sub>)<sub>h</sub>- where g = 1-2 and h = 1-2,

where i = 1 or 2, or

where j is 1 or 2 and Z is an amino group, alkylamino group, or piperidyl amino group; and

B is a substituted or unsubstituted C1 to C6 alkyl group;

W is a hydrogen or C(O)R where R is as defined above;

and pharmaceutically acceptable salts, esters or hydrates thereof.

## 13. A compound represented by formula I or II:

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II

#### 5 wherein

R is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, or heteroaryl group;

R<sup>2</sup> is -H or -CH<sub>3</sub>;

 $R^3$  is -H, -CH<sub>3</sub>, -CH<sub>2</sub>CONH<sub>2</sub> or -CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>;

10 R<sup>4</sup> is -H or -OH;

R<sup>5</sup> is -OH, -OPO<sub>3</sub>H<sub>2</sub>, or -OSO<sub>3</sub>H;

R<sup>6</sup> is -H or -OSO<sub>3</sub>H;

R<sup>7</sup> is -CH<sub>3</sub> or -H;

(Y) is represented by the following formula

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wherein

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A is -(CH<sub>2</sub>)<sub>a</sub>- where a = 1, 2 or 4, -CHR'-CHR"-(CH<sub>2</sub>)<sub>b</sub>- where R' and R" are independently -H, -OH,  $C_6H_5O$ -, -SH, -NH<sub>2</sub>,  $C_nH_{n+2}NH$ -,  $C_nH_{n+2}O$ -,  $C_nH_{n+2}S$ - or - $C_nH_{n+2}$  where n = 1-4 and b = 0, -(CH<sub>2</sub>)<sub>c</sub>- C(O)NH(CH<sub>2</sub>)<sub>d</sub>- where c = 1-2 and d = 1-2, -N=CH-(CH<sub>2</sub>)<sub>e</sub>- where e = 0-2, -NR"'(CH<sub>2</sub>)<sub>f</sub>- where R'" is -H, -C(O)CH<sub>2</sub>NH<sub>2</sub>, -C(O)CH(NH<sub>2</sub>)CH<sub>2</sub>NH<sub>2</sub> or - $C_nH_{n+2}$  where n = 1-4 and f = 1-3, -(CH<sub>2</sub>)<sub>g</sub>-SO<sub>2</sub>-(CH<sub>2</sub>)<sub>h</sub>- where g = 1-2 and h = 1-2,

where i = 1 or 2, or

where j is 1 or 2 and Z is an amino group, alkylamino group, or piperidyl amino group; and
B is a substituted or unsubstituted C1 to C6 alkyl group;
W is a hydrogen or C(O)R where R is as defined above;

and pharmaceutically acceptable salts, esters or hydrates thereof.

20 14. The compound of Claim 12 or 13 wherein R is a terphenyl group represented by the structure

15. A pharmaceutical composition comprising a compound of Claim 13 and a pharmaceutically inert carrier.

- 5 16. The pharmaceutical composition of Claim 15 further comprising a wetting agent, lubricating agent, emulsifier, suspending agent, preservative, sweetener, stabilizer, perfuming agent, flavoring agent or combinations thereof.
  - 17. A method of inhibiting fungal activity comprising contacting a compound of Claim 13 with a fungus.

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18. A method of treating a fungal infection in a human comprising administering to a human in need of such treatment a therapeutically effective amount of a compound according to Claim 13.

19. The method of Claim 18 wherein said compound is administered to said human topically, orally, by injection, by inhalation, or combinations thereof.

20. A method of inhibiting parasitic activity comprising contacting a compound of Claim 13 with a parasite.